

Experimental Characterization and First-Principles Understanding on the Structure of Zr-doped UO_2

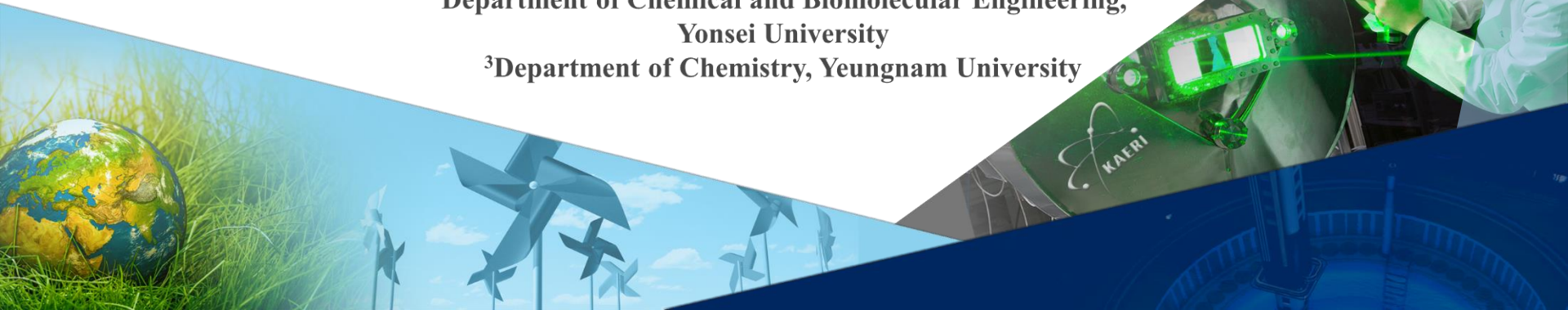
15th November 2019

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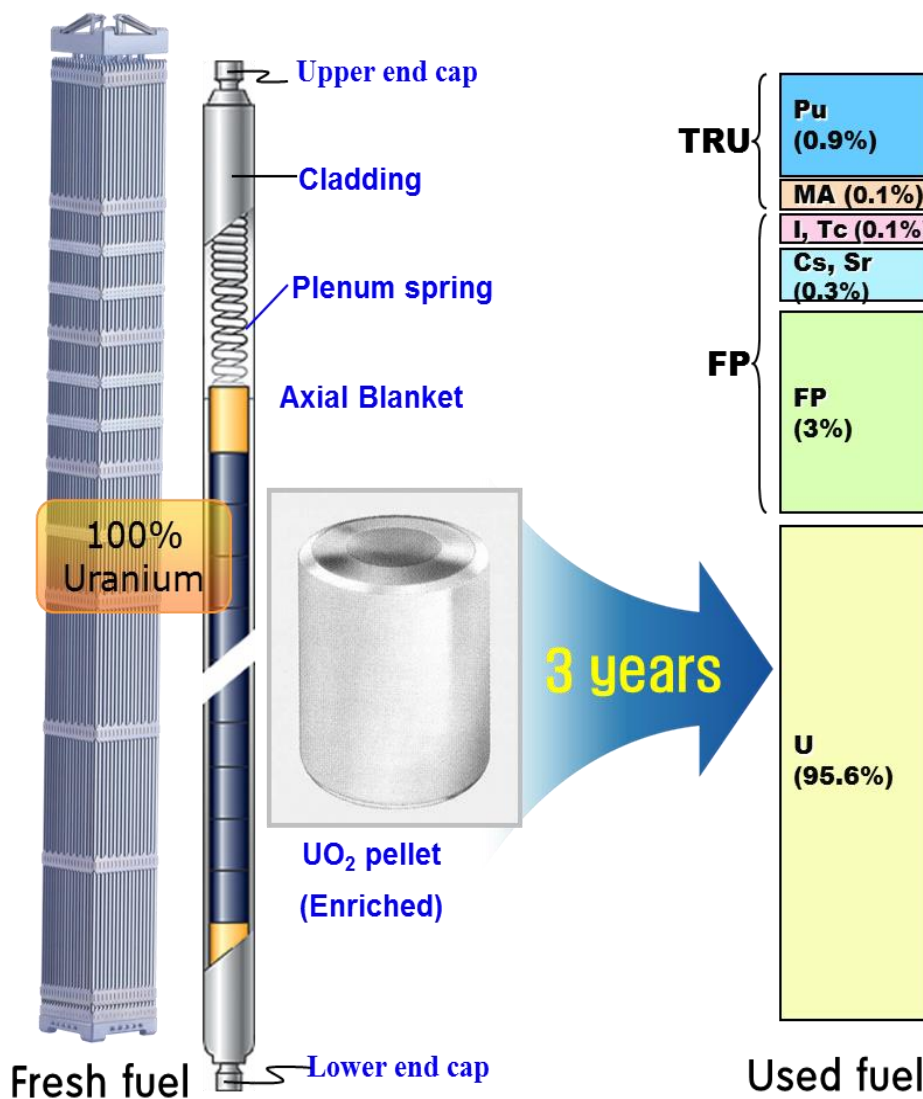
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I. Introduction



I. Introduction

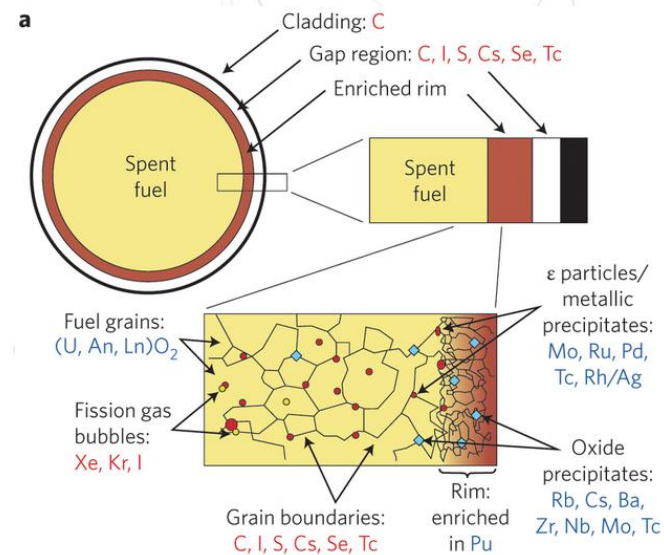
Distribution of fission products and actinides after irradiation



Chemical inventory for HBU SNF in CORE and OUT regions.

Element	HBU-CORE (μg/g)	HBU-OUT (μg/g)
Rb	500 ± 100	600 ± 120
Sr	800 ± 160	800 ± 160
Y	310 ± 130	340 ± 150
Zr	6300 ± 1300	7900 ± 1600
Mo	5900 ± 1180	8000 ± 1600
Tc	1300 ± 260	1600 ± 300
Ru	4400 ± 880	5300 ± 1100
Rh	650 ± 130	800 ± 160
Cs	3800 ± 200	4800 ± 240
La	2000 ± 400	2300 ± 500
Nd	7000 ± 1400	8000 ± 1600
U	780000 ± 78000	770000 ± 77000
Np	810 ± 160	760 ± 150
Pu	8000 ± 1600	11000 ± 2200
Am	840 ± 170	1200 ± 240
Cm	290 ± 60	460 ± 90

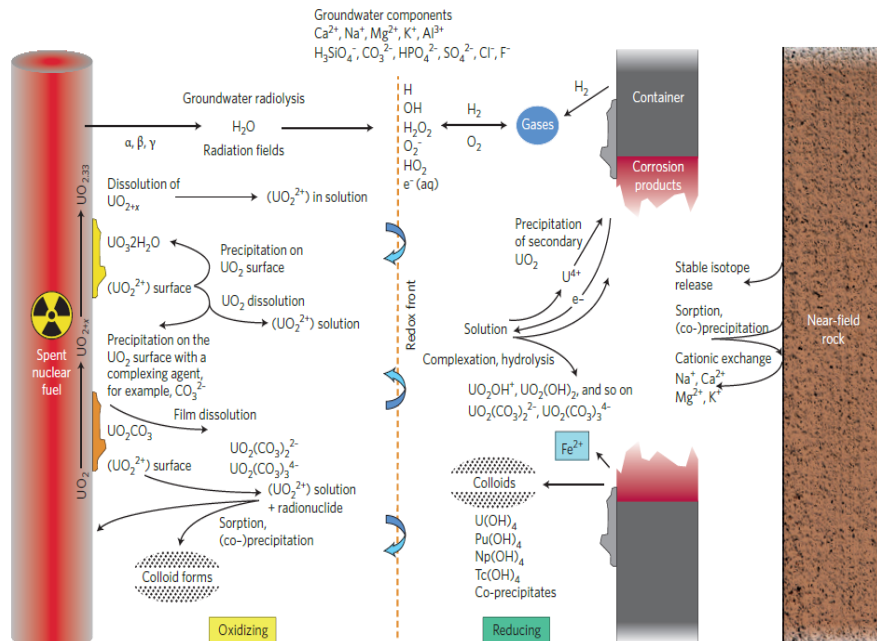
D. Serrano-Purroy *et al.*, J. Nucl. Mater. 427 (2012) 249-258



R.C. Ewing, Nature Mater. 14 (2015) 252-257

J. Bruno *et al.*, Elements 2 (2006) 343-349

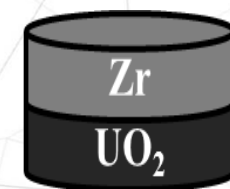
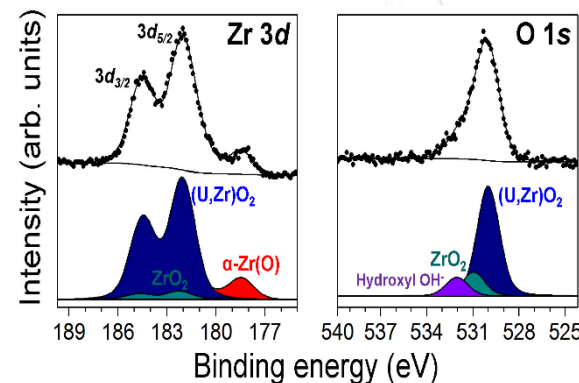
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UO₂-Zr sample

Y.-S. Youn *et al.*, Bull. Korean Chem. Soc. 36 (2015) 2068-2072

Component (fission products, actinides, burn-up)
Physical properties (structure, thermal conductivity)
Chemical Properties (chemical state, oxidation, reactivity)

Doping effect → **Structural change & Reactivity**

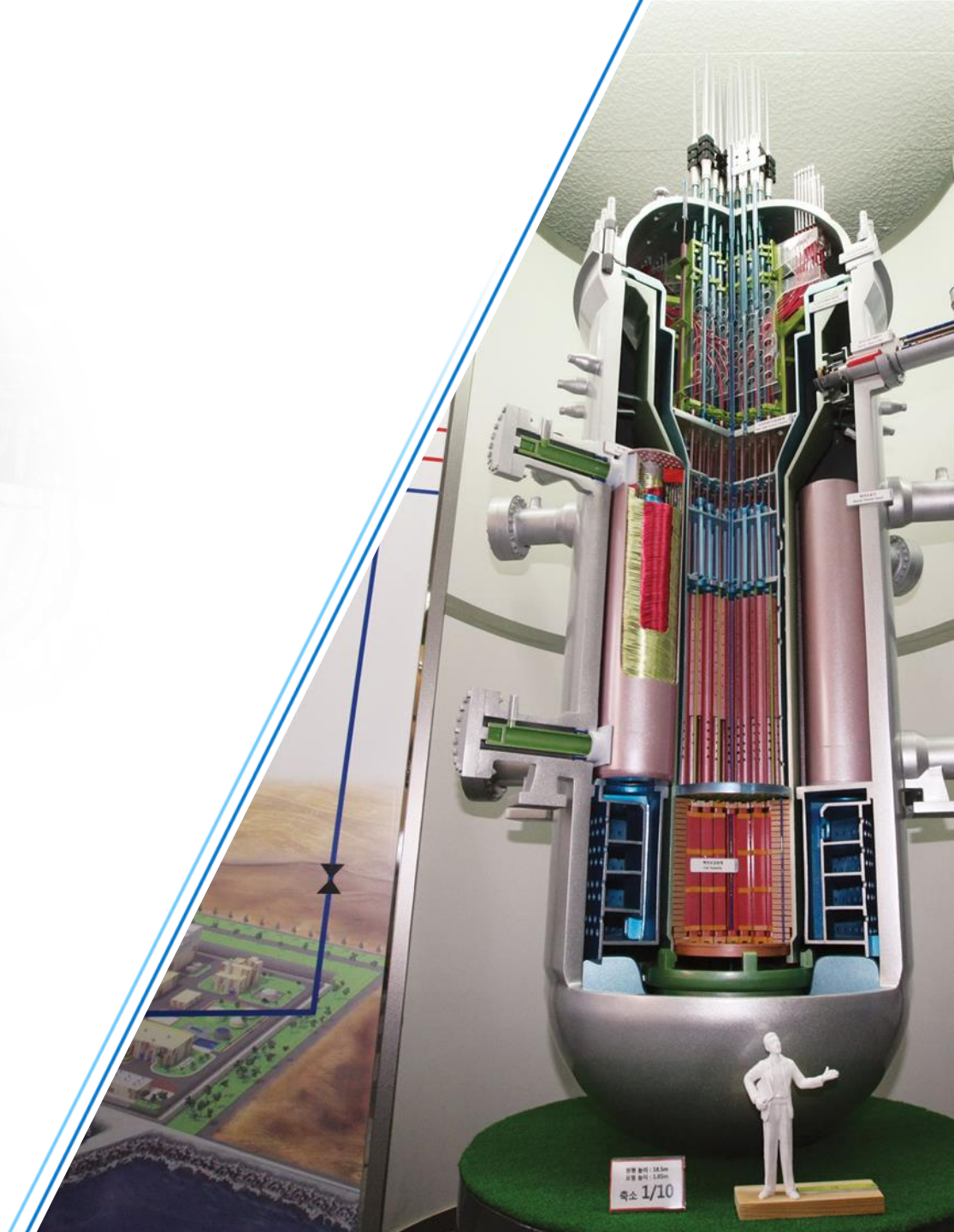
**Complex material,
Spent Nuclear Fuel**



**Simplified
single-element-doped UO₂**

Zr⁴⁺

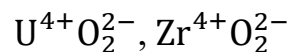
II. Experiment



II. Experiment

Preparation of pellet

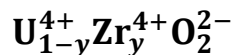
Powder mixing of UO_2 and ThO_2 powder



Pressed into a disk pellet



Sintering at 1700°C for 18 h in a reducing atmosphere
with flowing hydrogen

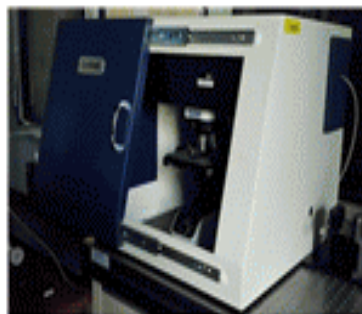


Thickness : < 1 mm
Diameter: ~ 9 mm

XRD (X-Ray Diffraction)

Bruker AXS D8 Advance X-ray Diffractometer using
 $\text{CuK}\alpha$ radiation at room temperature

Raman spectroscopy



ANDOR Shamrock SR500i spectrometer with a 632.8
nm excitation of He-Ne laser

First-principles DFT calculations

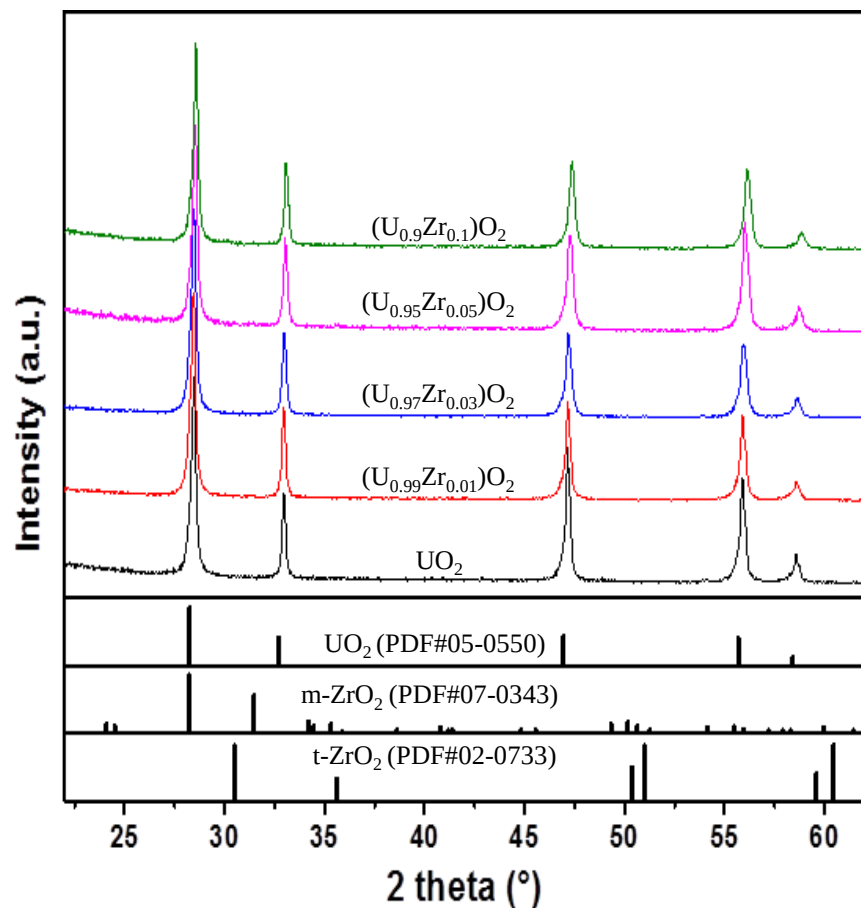
- DFT implemented in VASP
- The projector-augmented wave (PAW) method :
represent the core electrons
- The generalized gradient approximation (GGA) with
the Perdew-Burke-Ernzerhof (PBE):
correct the exchange and correlation potentials
- Energy cut-off of 600 eV were used for all calculations

III. Results & Discussion

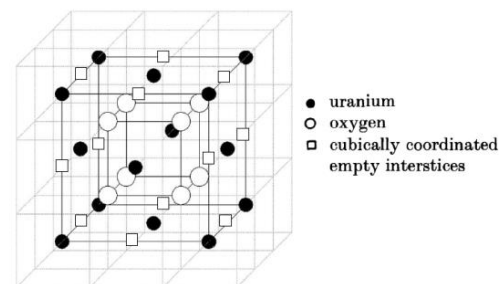


III. Results & Discussion

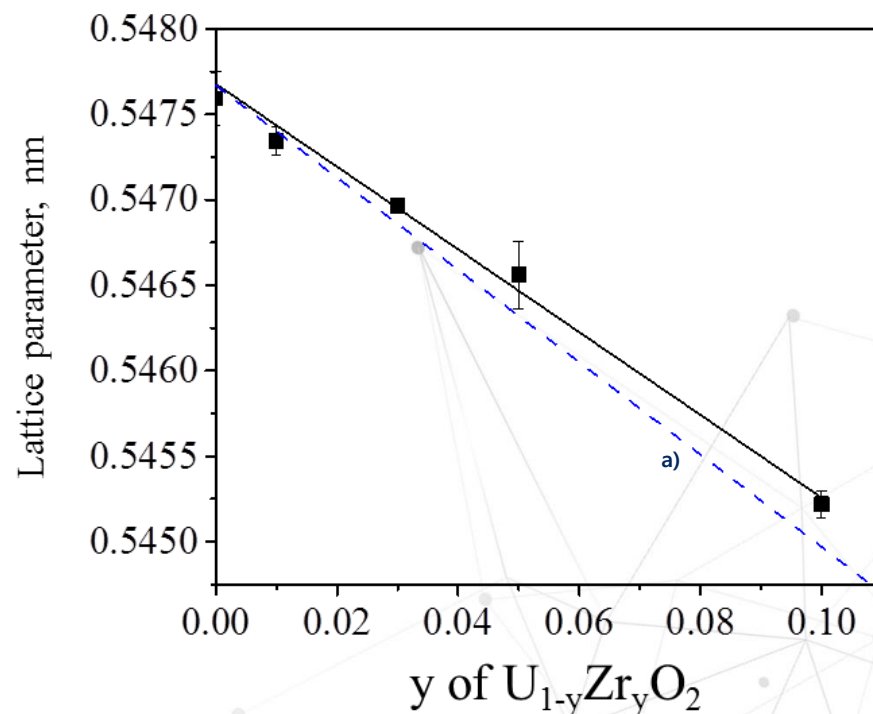
XRD data of (U,Zr)O₂



m : monoclinic, t : tetragonal



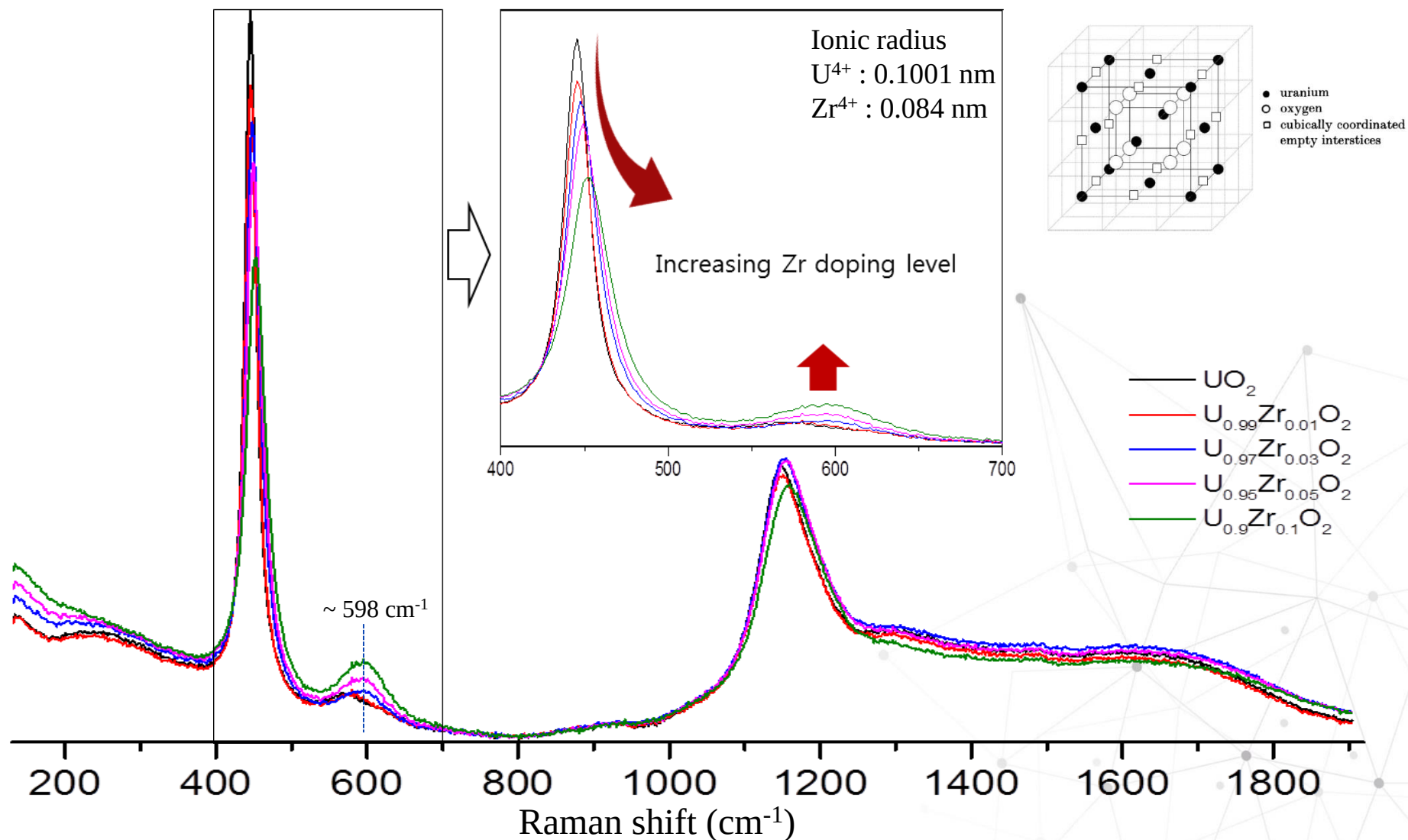
Ionic radius
 U^{4+} : 0.1001 nm
 Zr^{4+} : 0.084 nm



a) K. Une *et al.*, J. Am. Ceram. Soc. 66 (1983) C179-C180

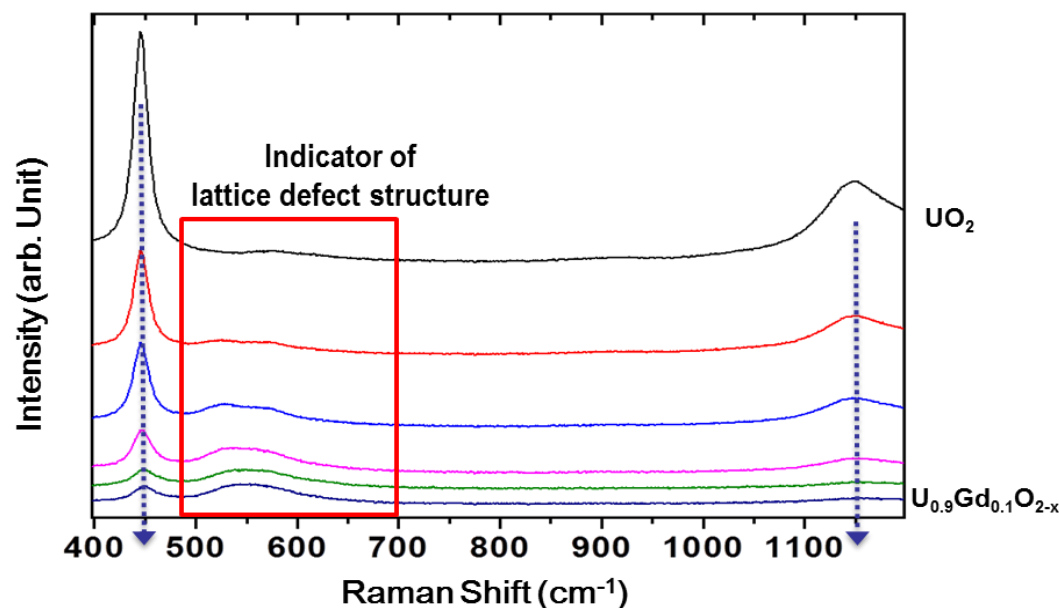
III. Results & Discussion

Raman spectra of (U,Zr)O₂

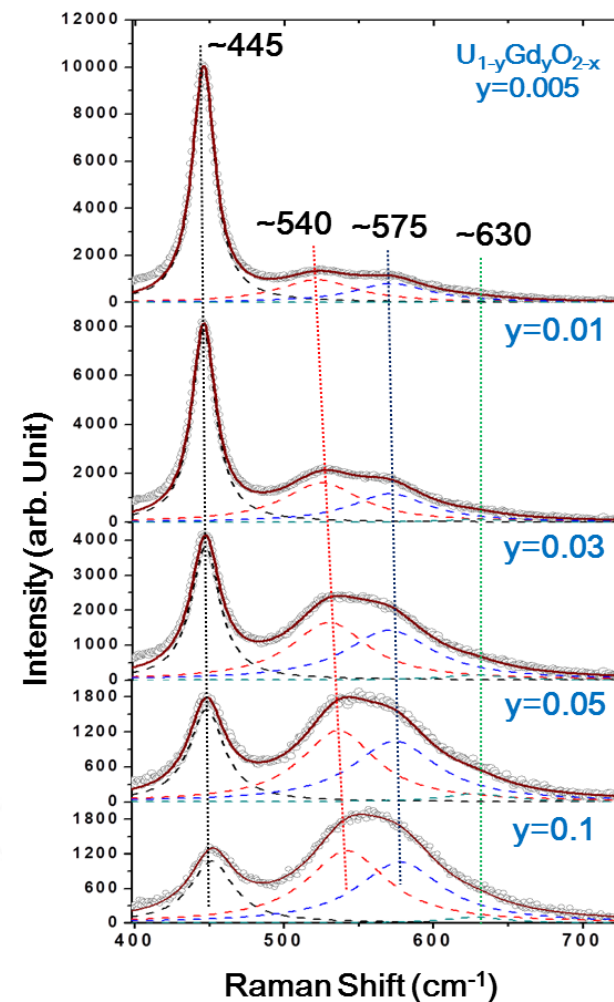


III. Results & Discussion

Raman spectra of (U,Gd)O₂

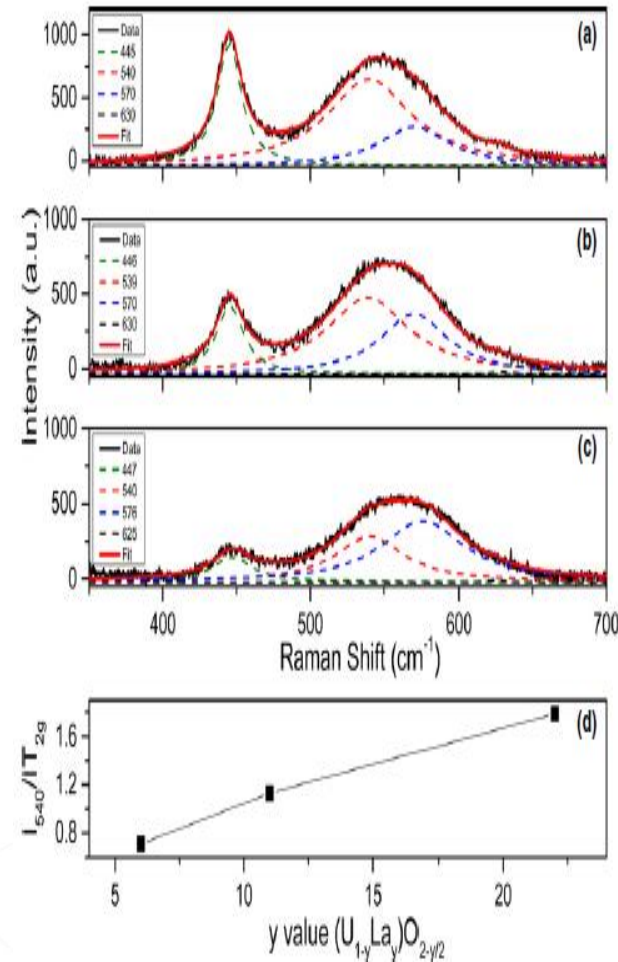
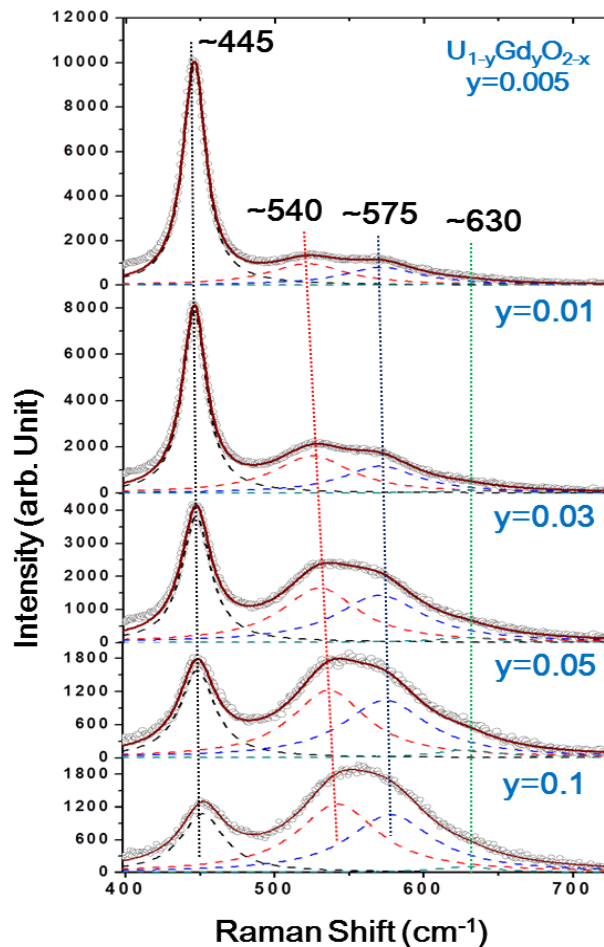
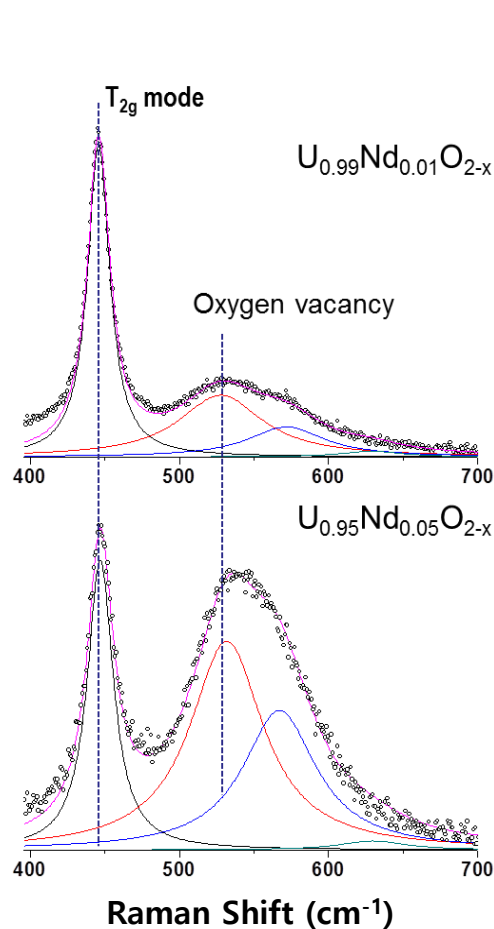


- ~445 cm⁻¹ : **T_{2g} fundamental O-U stretch** at perfect fluorite structure
- ~540 cm⁻¹ : Lattice distortion from **creation of oxygen vacancies** due to Gd³⁺
- ~575 cm⁻¹ : **L-O (Longitudinal-Optical) phonon** band
- ~630 cm⁻¹ : Cuboctahedral constituting of the **U₄O₉ phase**
- ~1150 cm⁻¹ : **2 L-O phonon**



III. Results & Discussion

Compare with other results

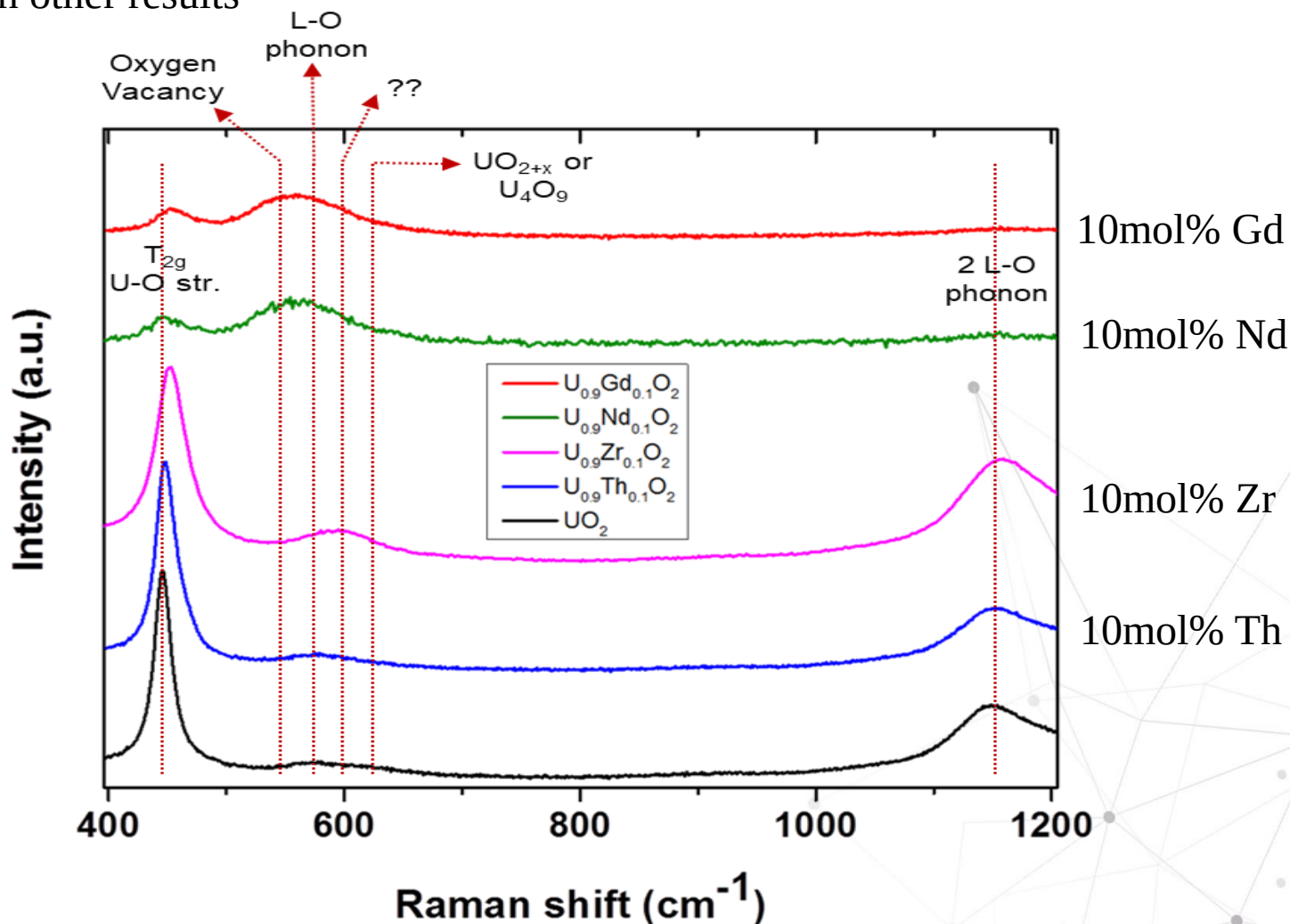


Ln³⁺ doping elements show similar features

$y = 0.06, 0.11, \text{ and } 0.22 \text{ in } U_{1-y}La_yO_{2-y/2}$

III. Results & Discussion

Compare with other results



J. Lee *et al.*, J. Nucl. Mater. 486 (2017) 216-221

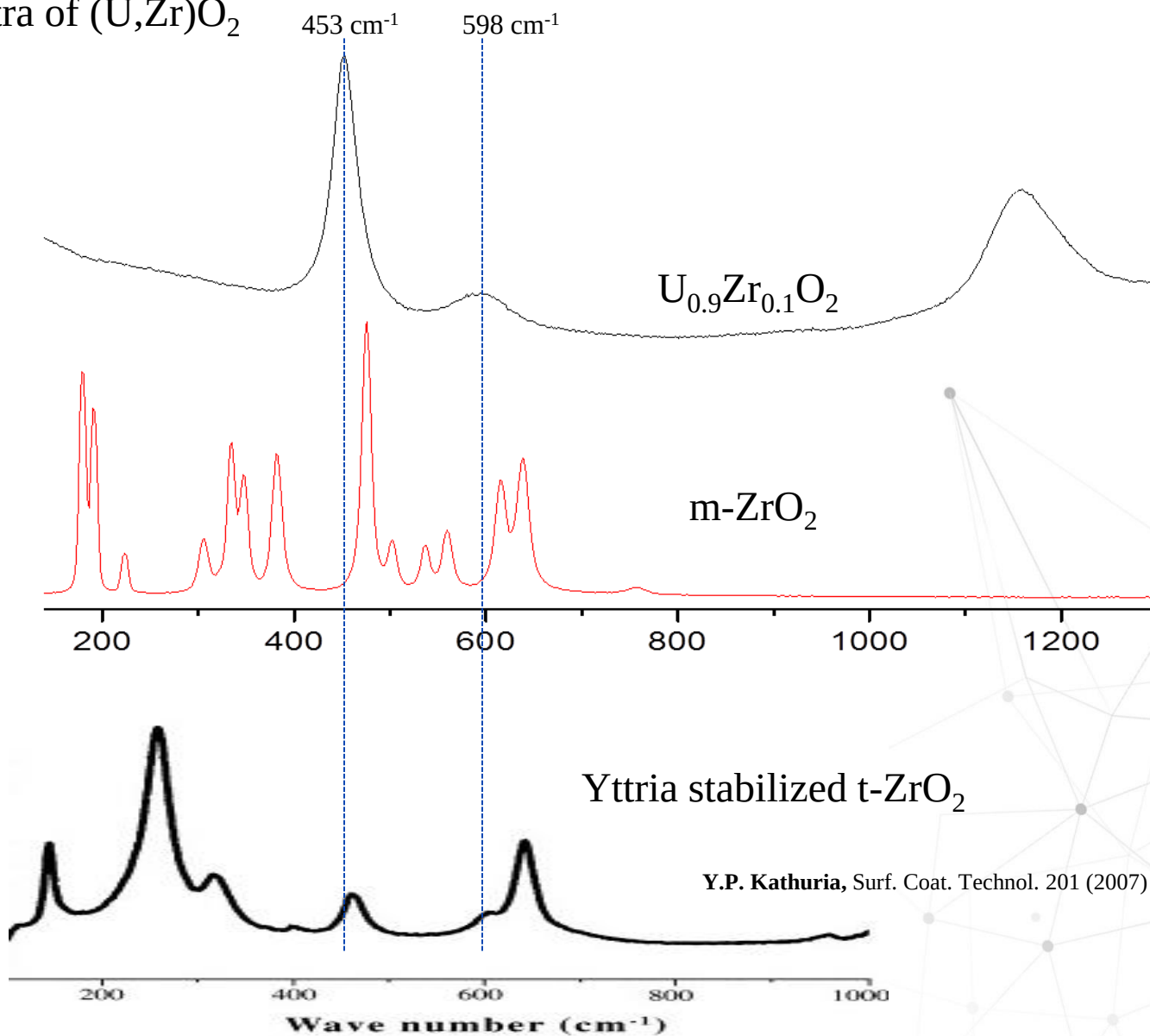
J. Lee *et al.*, Prog. Nucl. Sci. Technol. 5 (2018) 33-36

J. Lee *et al.*, Int. J. Energy Res. 43 (2019) 3322-2229

J. Lee *et al.*, J. Radioanal. Nucl. Chem. 316 (2018) 1295-1300

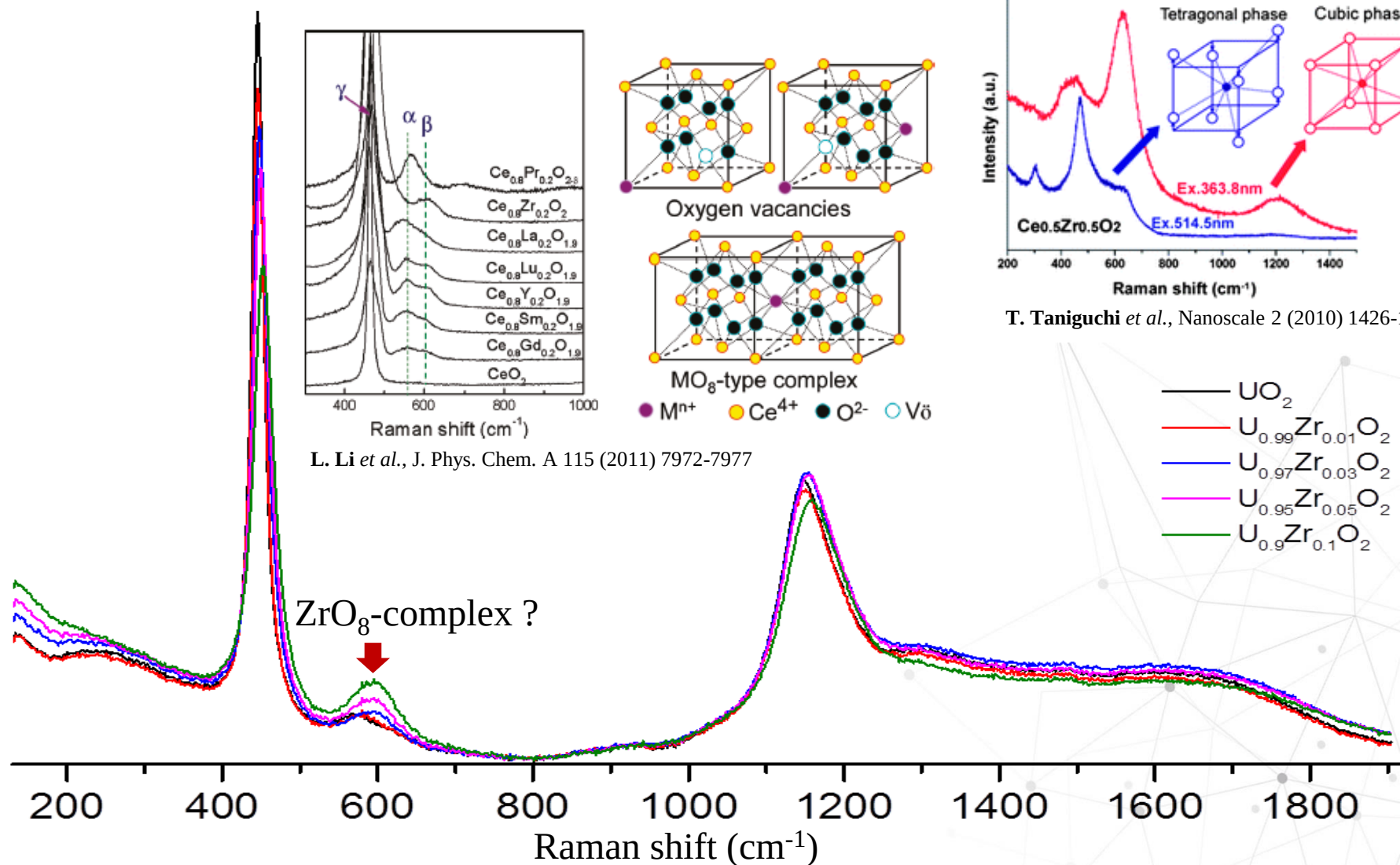
III. Results & Discussion

Raman spectra of (U,Zr)O₂



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Raman spectra of (U,Zr)O₂



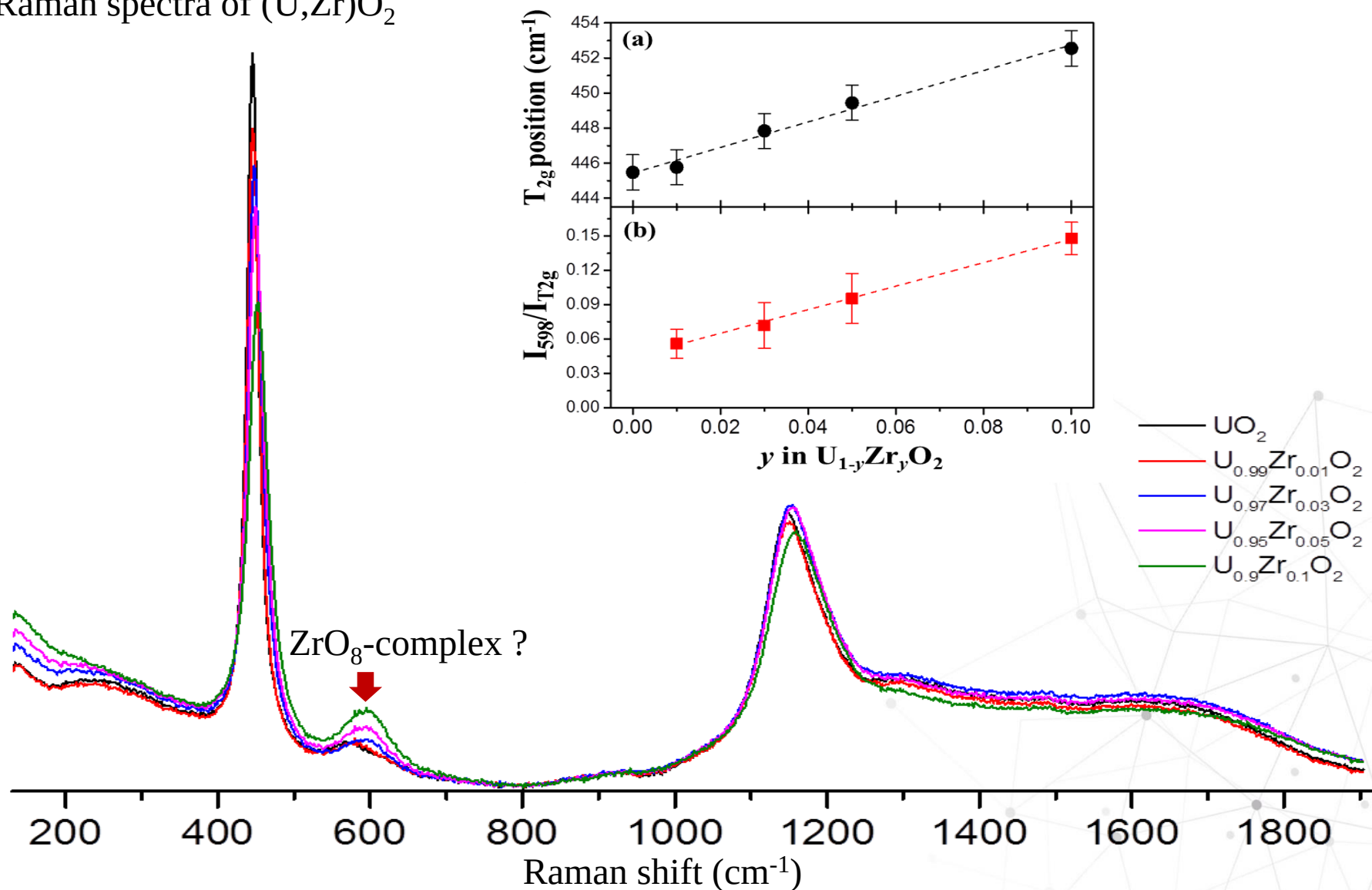
L. Li *et al.*, J. Phys. Chem. A 115 (2011) 7972-7977

T. Taniguchi *et al.*, Nanoscale 2 (2010) 1426-1428

J. Lee *et al.*, Int. J. Energy Res. 43 (2019) 3322-2229

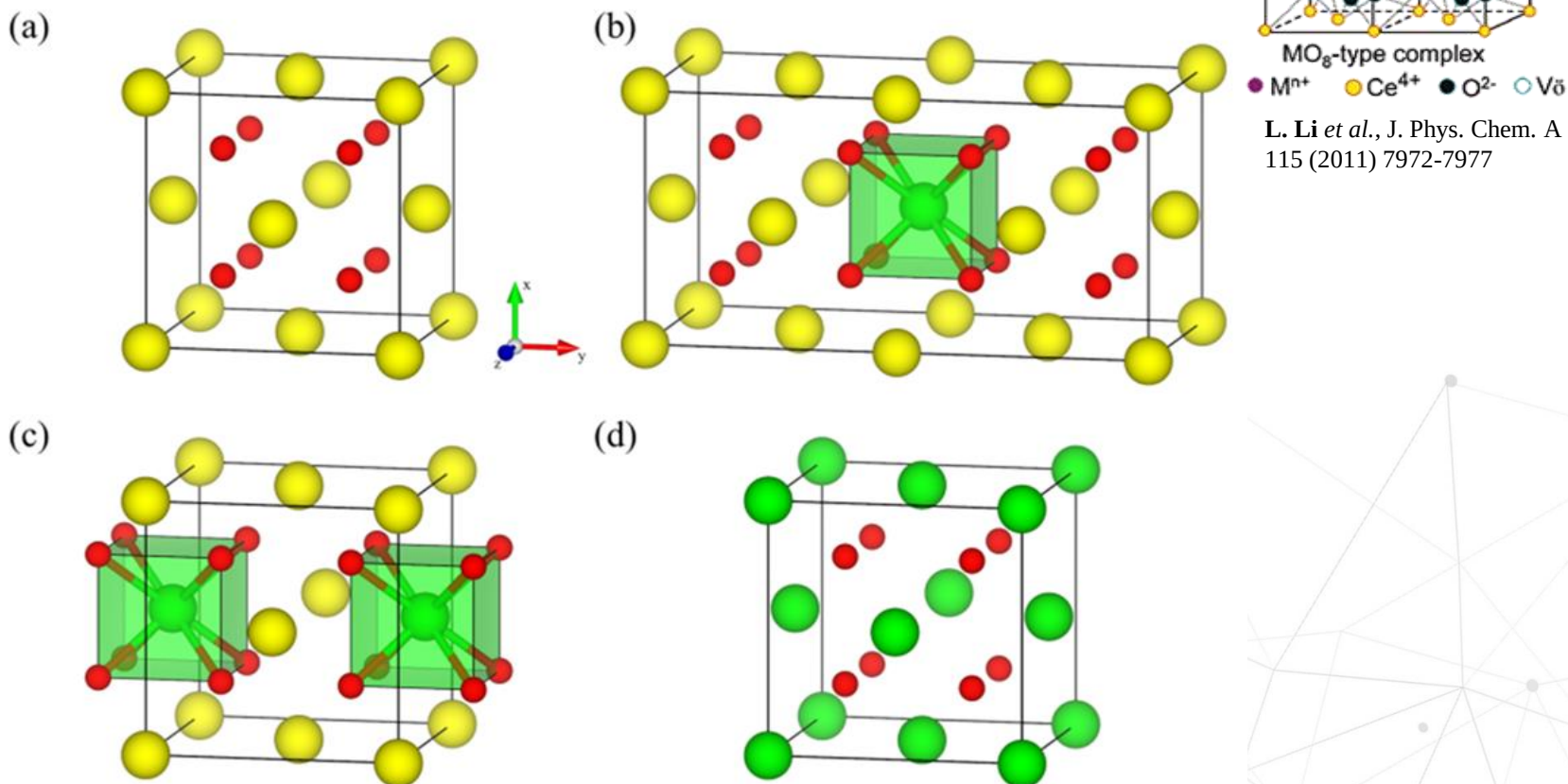
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Raman spectra of (U,Zr)O₂



III. Results & Discussion

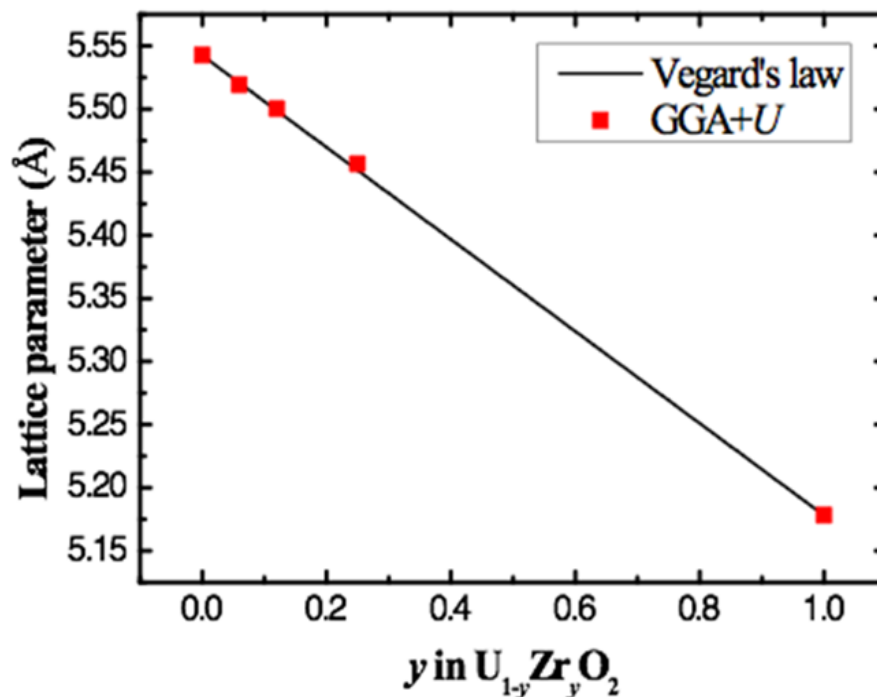
First-principles DFT calculations



Model systems of (a) UO_2 , (b) $\text{U}_{0.87}\text{Zr}_{0.13}\text{O}_2$, (c) $\text{U}_{0.75}\text{Zr}_{0.25}\text{O}_2$, and (d) cubic ZrO_2 . The yellow, green, and red spheres represent U, Zr, and O atoms, respectively

III. Results & Discussion

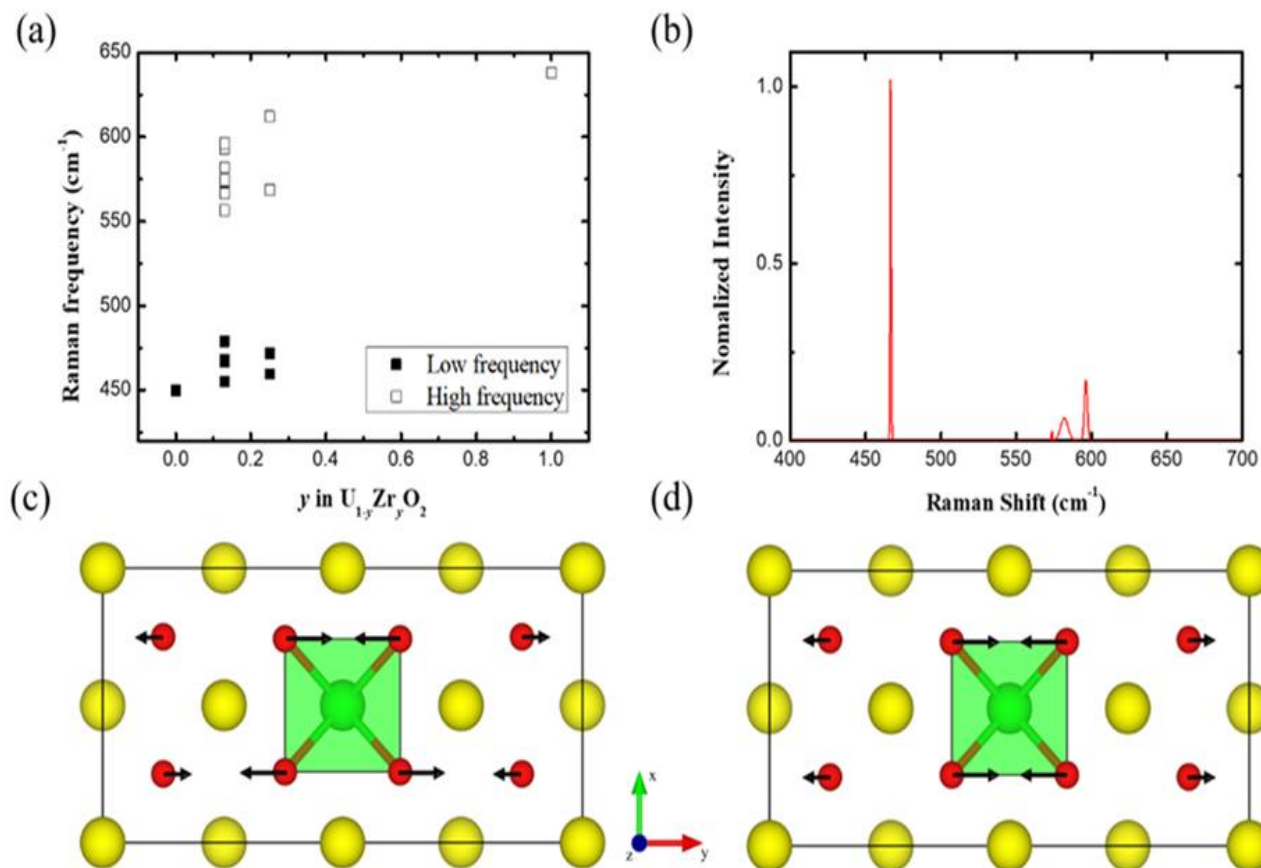
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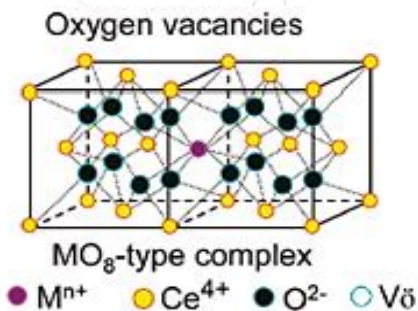
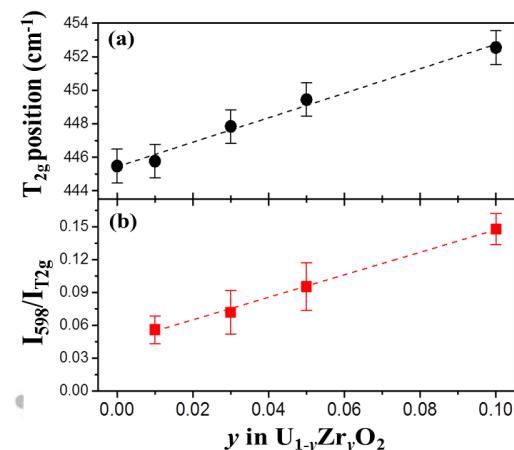
Relationship between the simulated lattice parameter (red squares) calculated using GGA+ U and Zr content (y) in $U_{1-y}Zr_yO_2$ ($y = 0.0, 0.06, 0.13, 0.25$, and 1). This linear relationship obeys Vegard's law (solid black line).

III. Results & Discussion

First-principles DFT calculations



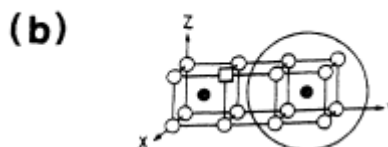
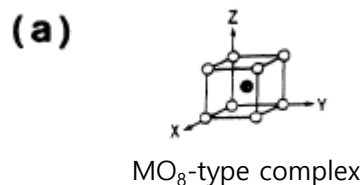
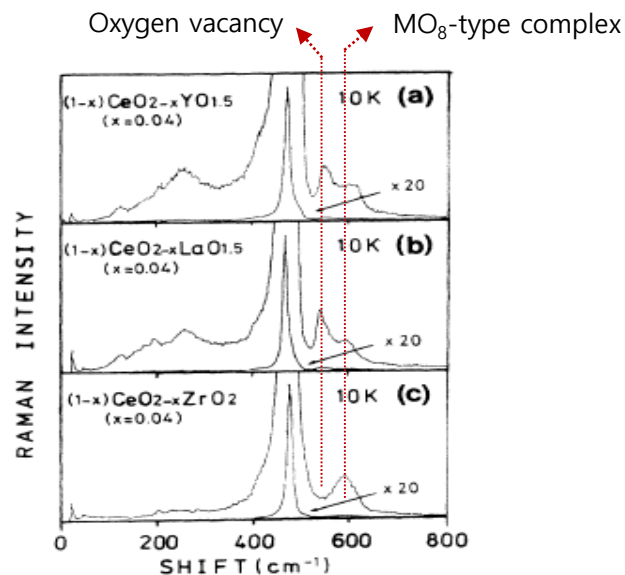
(a) Vibrational frequencies of $U_{1-y}Zr_yO_2$ as a function of Zr composition (y). (b) The simulated intensities and Raman frequencies of $U_{0.87}Zr_{0.13}O_2$. Normal modes of the peaks at (c) 466.82 cm^{-1} and (d) 596.15 cm^{-1} . The yellow, green and red spheres represent U, Zr and O atoms, respectively.



Ionic radius
 U^{4+} : 0.1001 nm
 Zr^{4+} : 0.084 nm

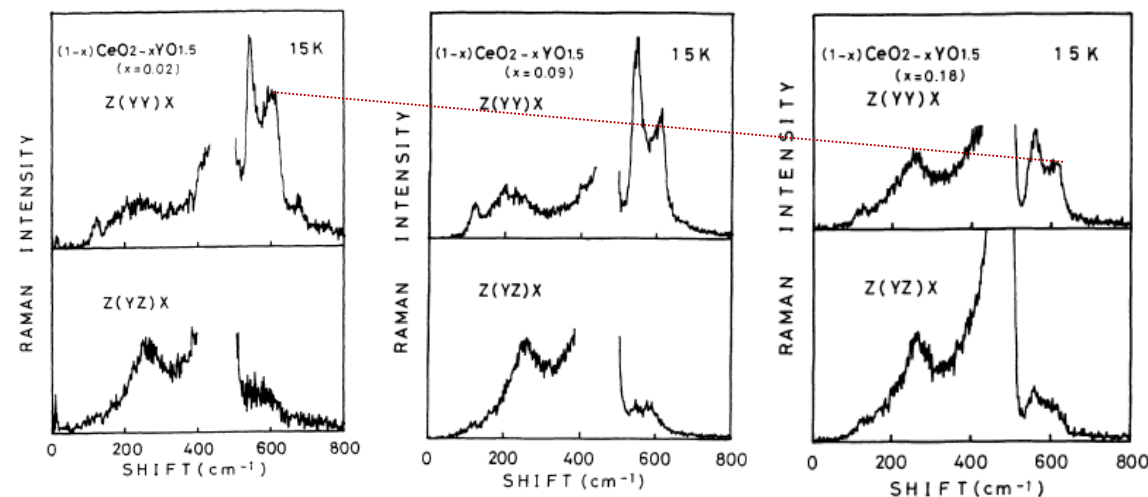
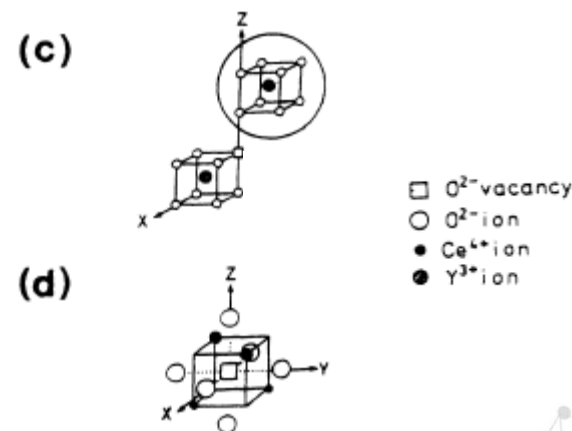
III. Results & Discussion

How about M(III)O₈-type complex?



(b), (c), (d) Defect spaces related to oxygen vacancy

(Ce,Y)O₂



The existence of a **MO₈-type** complex :

Dopant M is in 8-fold coordination with O²⁻ (O_h symmetry) excluding oxygen vacancy

Increasing the concentration of Ov :

the space differs from the MO₈-type complex since **the space no longer has O_h symmetry** due to the effect of the neighboring **oxygen vacancy**.

IV. Summary



IV. Summary

Summary

- The surface structure of $U_{1-y}Zr_yO_2$ pellets have been studied by XRD and Raman spectroscopy
- There is no defect structure related to oxygen vacancy
- The peak at $\sim 598\text{ cm}^{-1}$ in $U_{1-y}Zr_yO_2$ pellet is supposed to be the existence of ZrO_8 -type complex in UO_2 matrix.
- T_{2g} Raman vibration peak around 445 cm^{-1} shows positive shift and the I_{598}/I_{T2g} linearly increases with increasing Zr doping level
- XRD analysis and Raman spectroscopic observation was consistently supported by first-principles DFT calculation (O in the ZrO_8 -type complex moves longer distance)
- This study could provide important implication for predicting physicochemical properties of a corium (the mixture of core materials, UO_2 and ZrO_2) and reaction product of fuel and cladding

Ongoing and future work

- Electrochemical experiments with $(U,Zr)O_2$
- Experimental characterization integrated with first-principles computational validation on the surface structure of $(U,M^{3+})O_2$



Thank you!