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Synthesis, Characterization and Corrosion of Simulant Nuclear Fuel Debris

Dr. Claire Corkhill

30th Spent Nuclear Fuel Workshop, Gent, 15th November 2019.



@clairecorkhill
@ISL_Sheffield

Acknowledgements



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NDA
Nuclear
Decommissioning
Authority

Clemence Gause, Daniel Bailey, Lucy Mottram, Hao Ding, Liam Harnett, Sean Barlow, Martin Stennett, Adam Fisher, Neil Hyatt.



University of
BRISTOL

Haris Paraskevoulakos,
Mahmoud Mostafavi



東京大学
THE UNIVERSITY OF TOKYO

Mituru Uesaka,
Yuki Mitsuya



Tadhiro Washiya
Tomooki Shiba

PAUL SCHERRER INSTITUT
PSI

Daniel Grolimund,
Peter Warnicke



Viktor Krasnov



Sergey Sayenko



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CRP on Management of
Severely Damaged
Spent Fuel

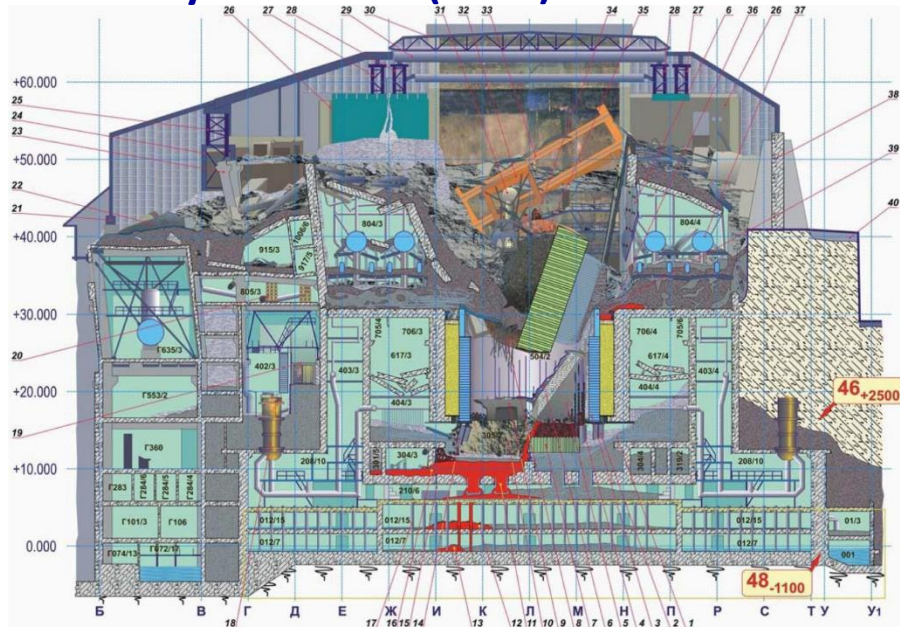
Decommissioning a nuclear “meltdown”



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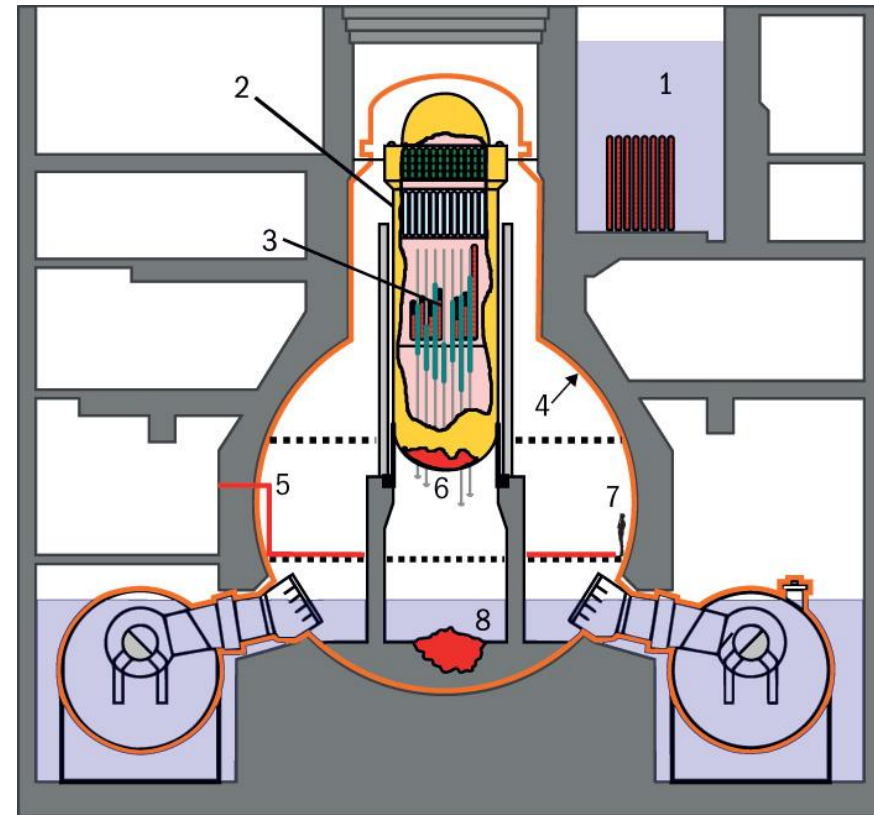
Chernobyl Reactor 4 (1986)



1. **How to remove fuel?** (mechanical properties, hardness etc.)
2. **Radioactive dust generation?** (respirable hazard)

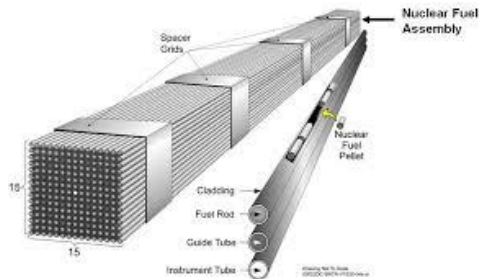
**Real materials are scarce (or not available)
and difficult to handle.**

Fukushima (2011)



- | | |
|-------------------------------------|---|
| 1: Spent fuel pool | 5: Path of Hitachi shape-changing robot |
| 2: Reactor pressure vessel (RPV) | 6: Damaged control rod drives |
| 3: Damaged core assembly | 7: Person for scale |
| 4: Primary containment vessel (PCV) | 8: Fuel debris/corium on basement floor |

Nuclear Fuel Debris

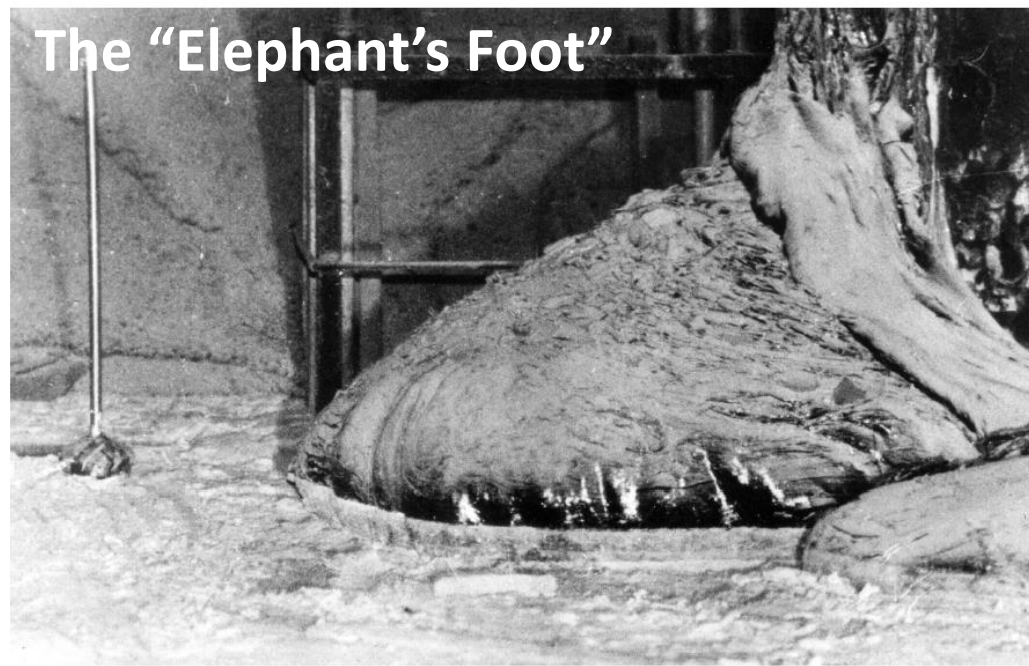


$U_{1-x}Zr_xO_2$
solid solution



$U_{1-x}Zr_xO_2$ +
Concrete
+ Stainless steel

LFCM (Lava-like Fuel Containing Material)
MCCI (Molten Core-Concrete Interaction)



Brown Lava



Black Lava

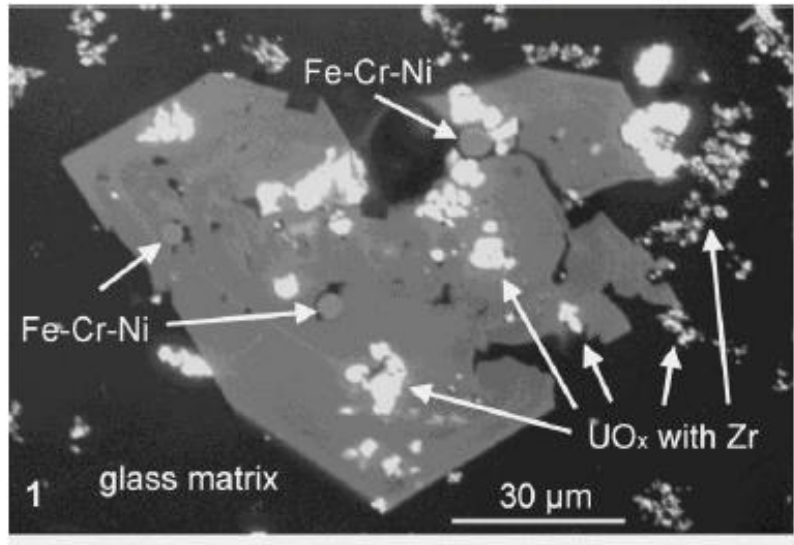
Pictures courtesy of Boris Burakov.

Lava-like fuel containing material

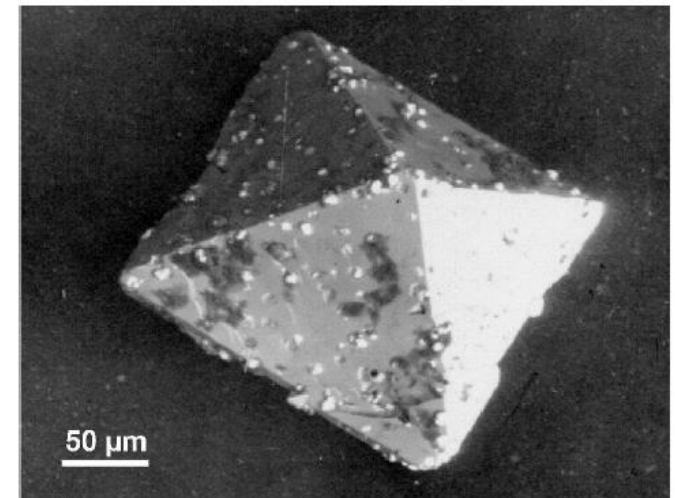


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- **Cubic UO_x** : similar to stoichiometric UO_2 ($a = 5.462\text{--}5.473 \text{ \AA}$)
- **Cubic UO_x with Zr**: $(\text{U}_{0.985}\text{Zr}_{0.015})\text{O}_2$ to $(\text{U}_{0.895}\text{Zr}_{0.105})\text{O}_2$ (a decreases from 5.468 to 5.318 $\text{\AA}$)
- **Tetragonal phase Zr-U-O** with varied chemical composition from $(\text{Zr}_{0.86}\text{U}_{0.14})\text{O}_2$ to $(\text{Zr}_{0.89}\text{U}_{0.11})\text{O}_2$
- **Monoclinic zirconia** with U (up to 6 wt.% U) with varied chemical composition from $(\text{Zr}_{0.995}\text{U}_{0.005})\text{O}_2$ to $(\text{Zr}_{0.967}\text{U}_{0.033})\text{O}_2$
- **High-uranium zircon (chernobylite)** $(\text{Zr}_{0.95}\text{U}_{0.05})\text{SiO}_4$ to $(\text{Zr}_{0.90}\text{U}_{0.10})\text{SiO}_4$ ($a = 6.617$; $c = 5.990 \text{ \AA}$).
- Little known about the glass matrix

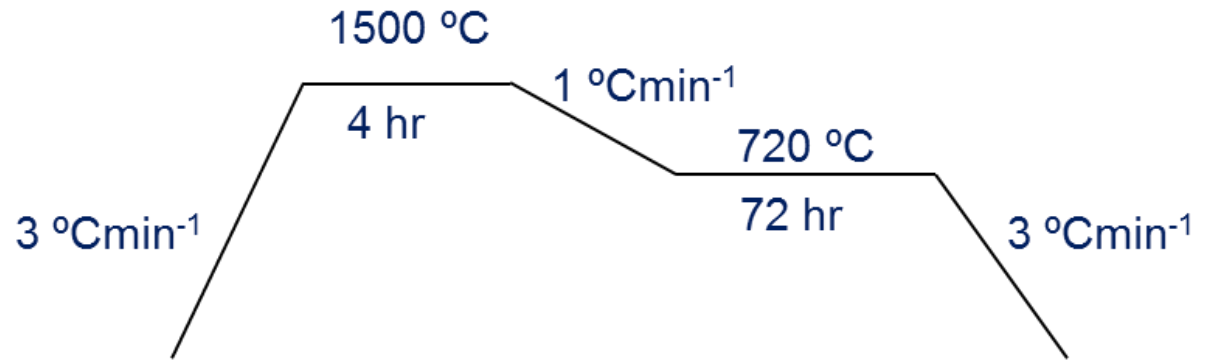


Chernobyl LFCM synthesis



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- Material batched according to compositions reported in literature
- Batches melted in alumina crucibles under H₂/N₂ atmosphere- prevent oxidation of U
- Dual-purpose annealing step- reduce fragility of samples & increase crystallinity
- Initial batches did not form ZrSiO₄
- Batches reformulated with increased Zr content → ZrSiO₄ formed

Chernobyl LFCM synthesis



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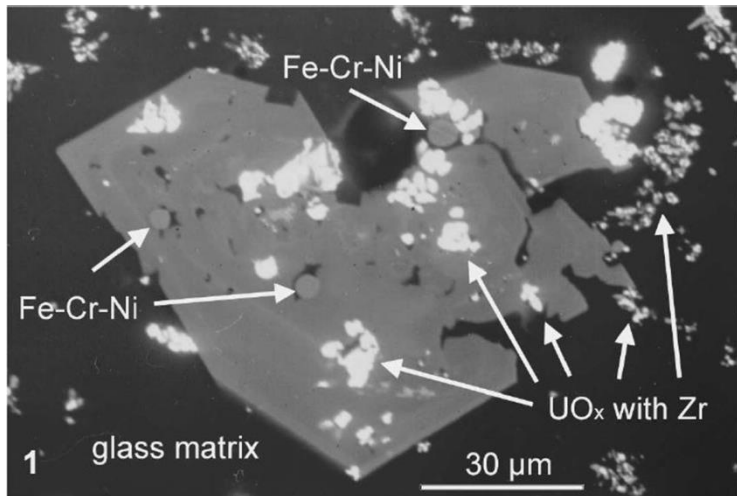
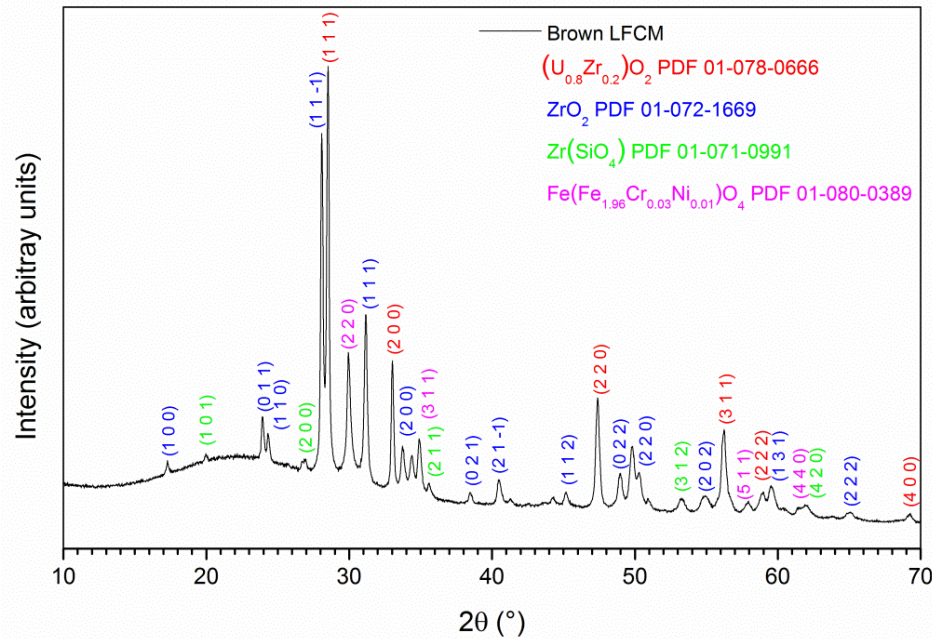
Brown lava: Iron contamination and high uranium content (~10 wt%)

Component	Mol%	Wt%
SiO ₂	65.2	58.0
CaO	7.3	6.3
ZrO ₂	7.6	14.6
Na ₂ O	7.9	7.6
Al ₂ O ₃	1.9	3.0
Fe ₂ O ₃ (Stainless Steel)	0.6	1.4
MgO	10.1	6.4
UO ₂	2.4	10.0

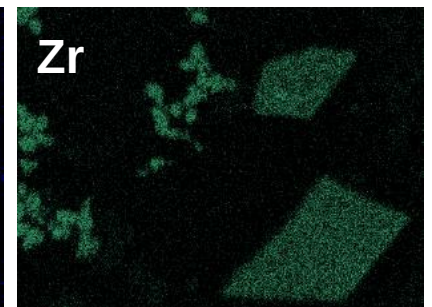
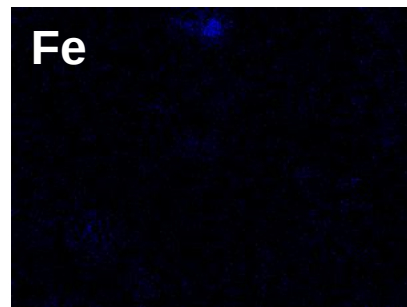
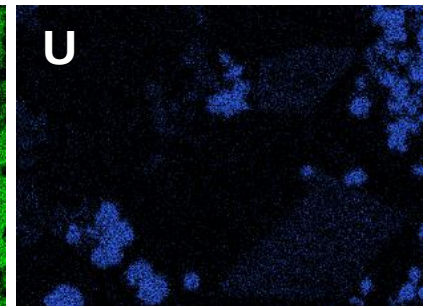
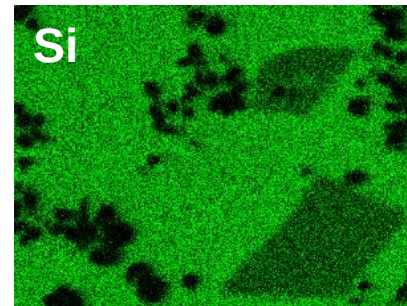
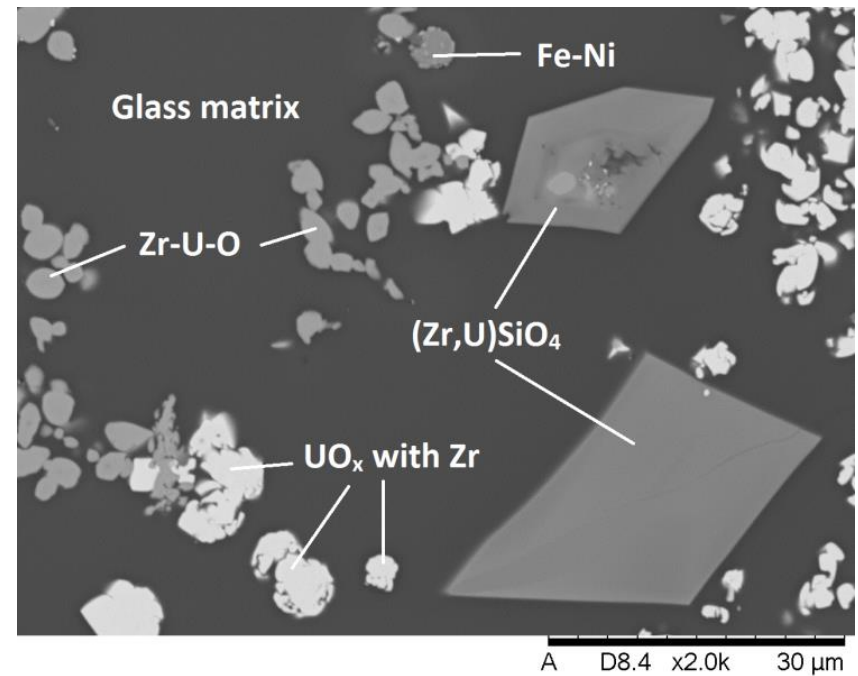


Brown LFCM

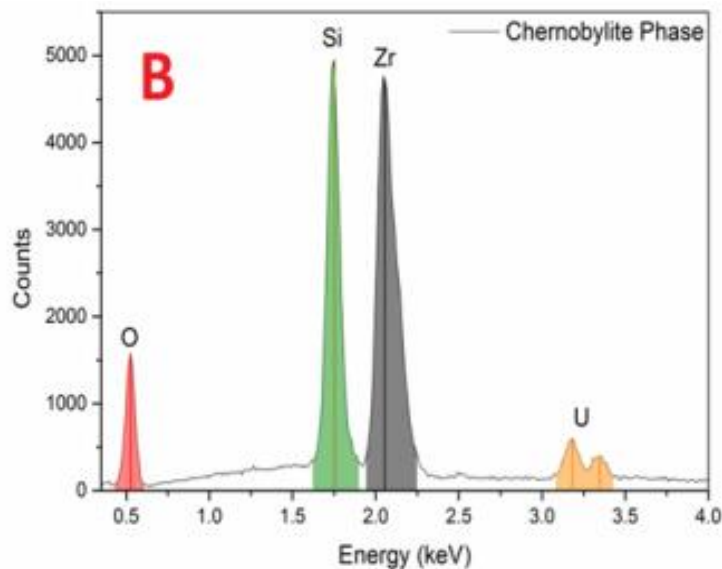
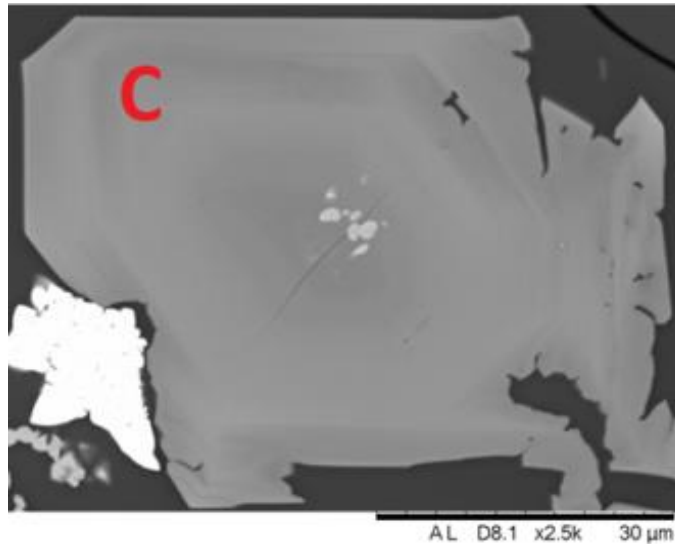
Simulant **Brown** LFCM



“Real”
LFCM by
Boris
Burakov

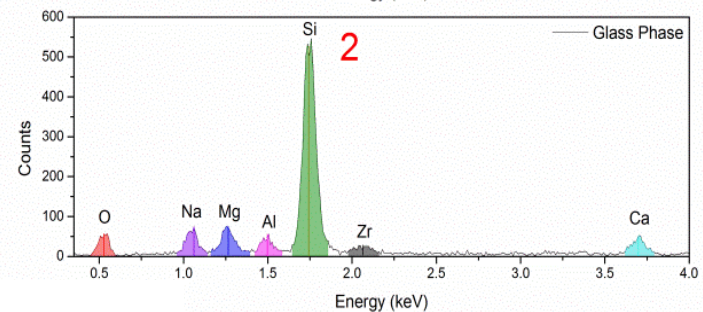
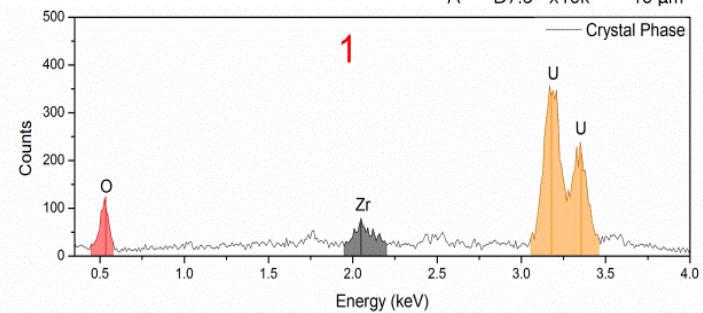
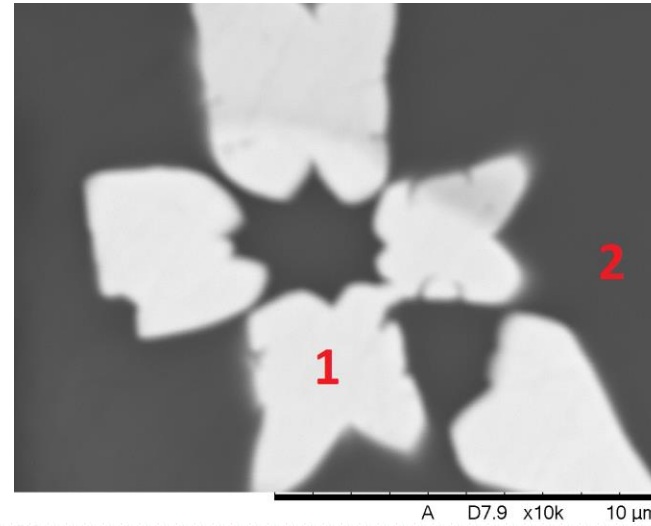


Simulant **Brown** LFCM



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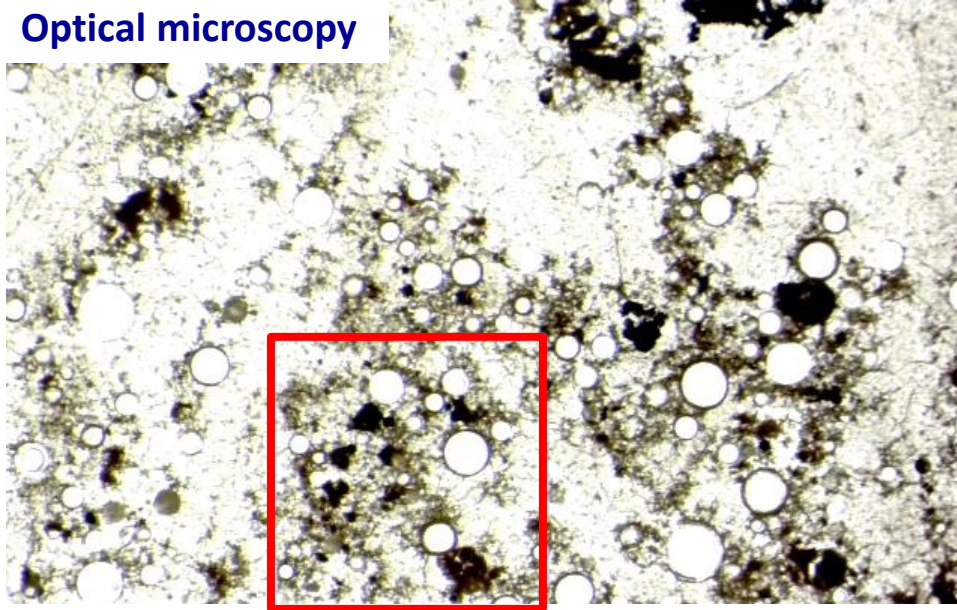
Simulant **Brown** LFCM: Advanced characterisation



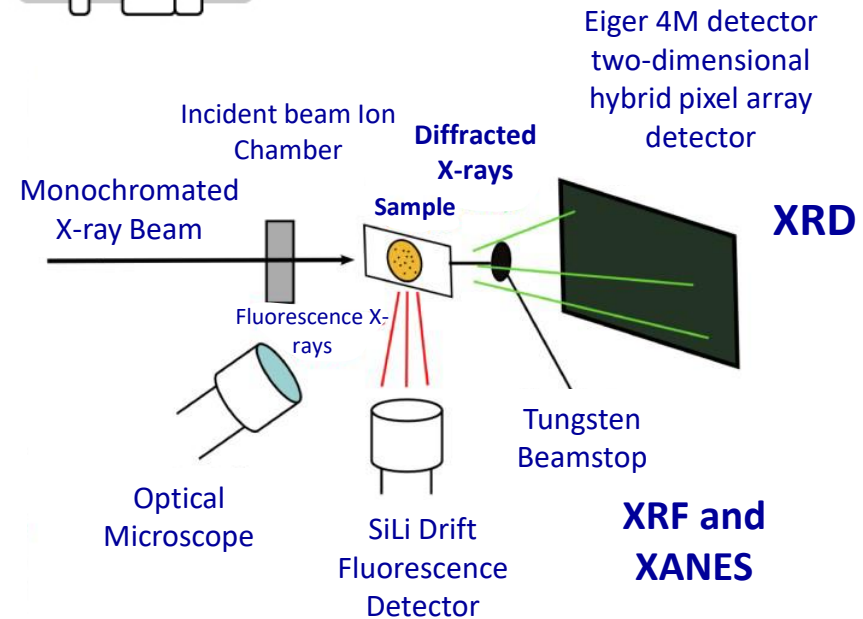
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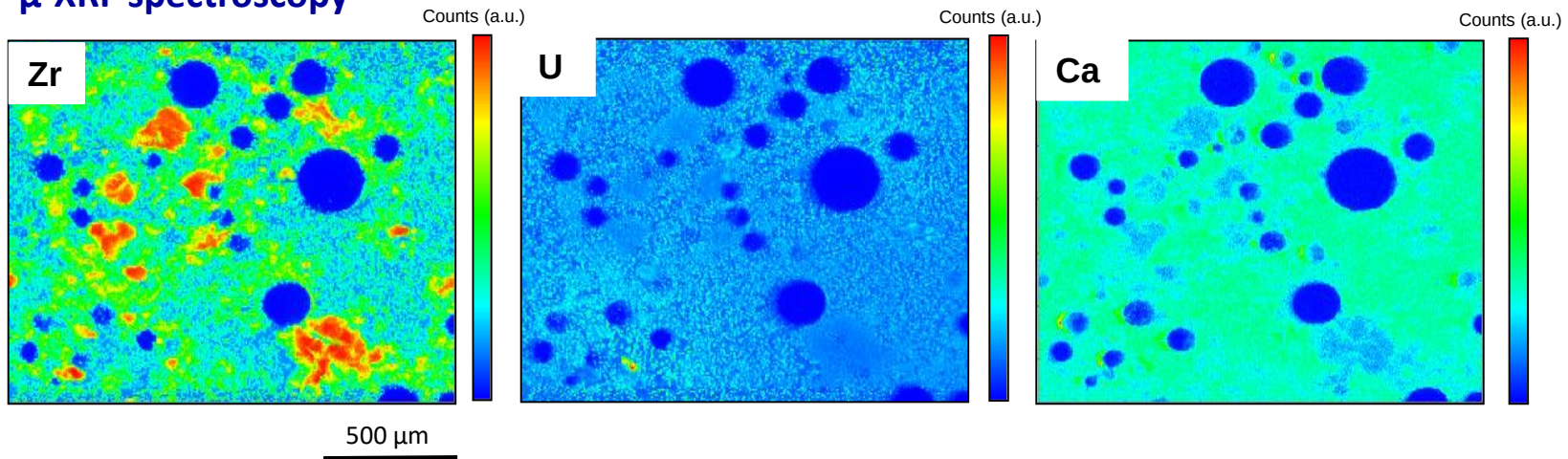
Optical microscopy



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μ -XRF spectroscopy



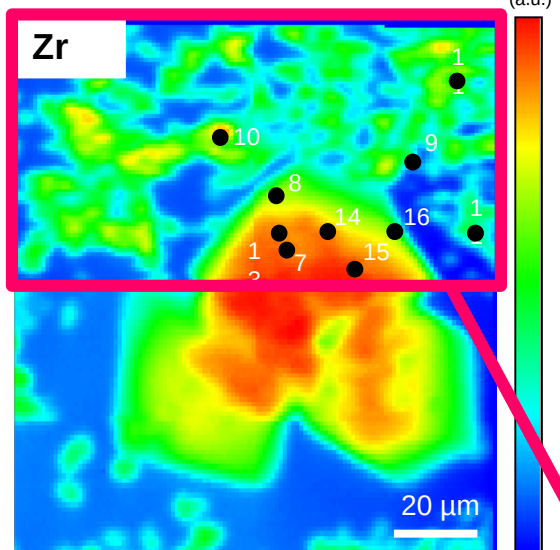
Simulant **Brown** LFCM: Advanced characterisation



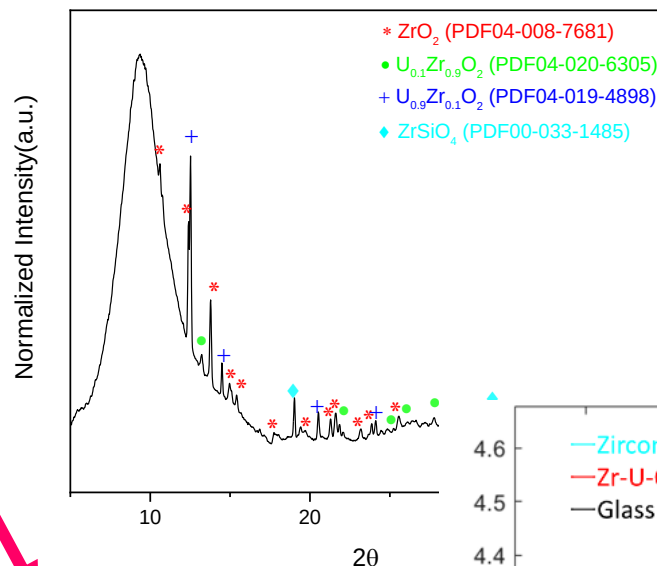
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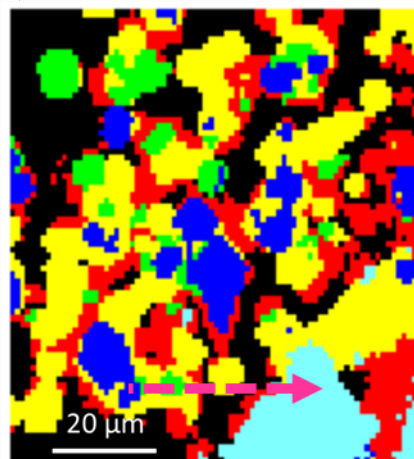
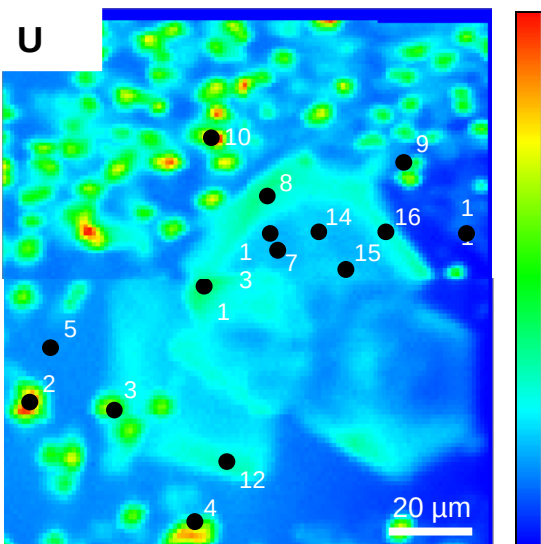
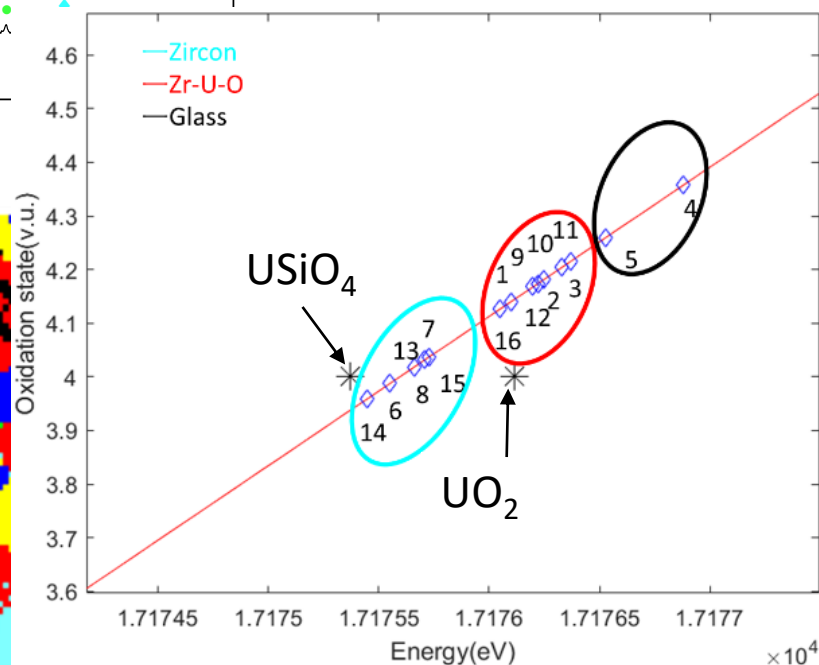
μ -XRF spectroscopy



μ -XRD mapping: phase identification



U L_{III} -edge μ -XANES

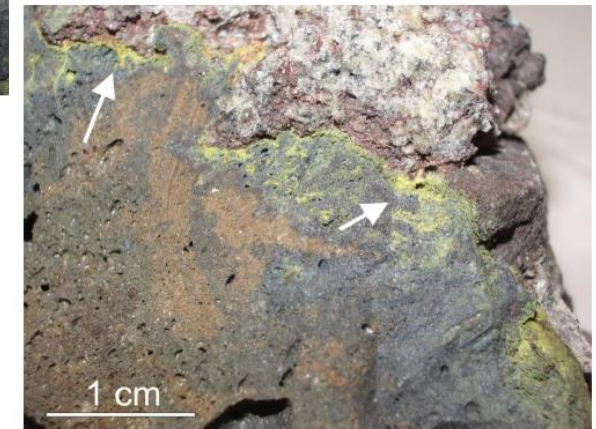
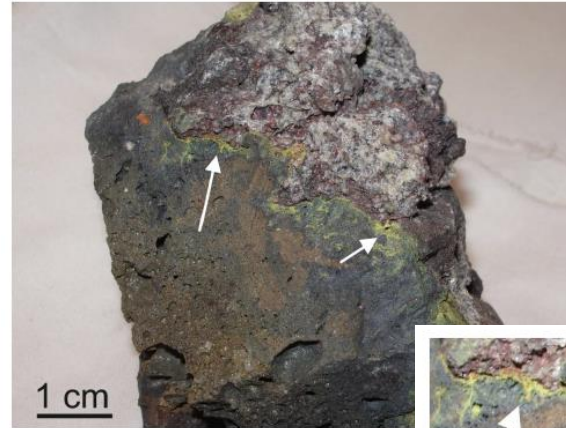


Nuclear fuel debris corrosion



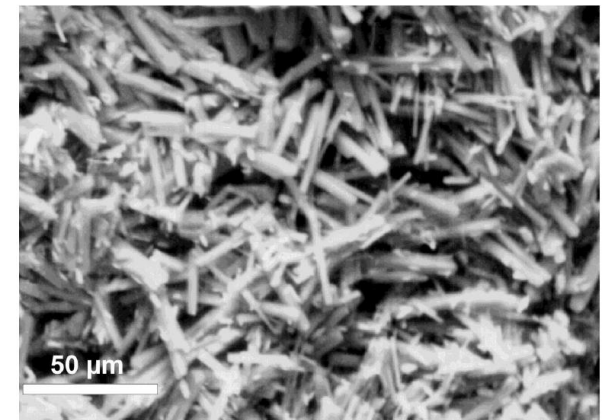
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Evidence of secondary mineralisation in Chernobyl reactor, and samples removed to laboratory

- $\text{Na}_3\text{H}(\text{CO}_3)_2 \cdot 2\text{H}_2\text{O}$
- $\text{UO}_4 \cdot 4\text{H}_2\text{O}$ (studtite)
- $\text{UO}_3 \cdot 2\text{H}_2\text{O}$
- UO_2CO_3
- $\text{Na}_4(\text{UO}_2)(\text{CO}_3)_3$

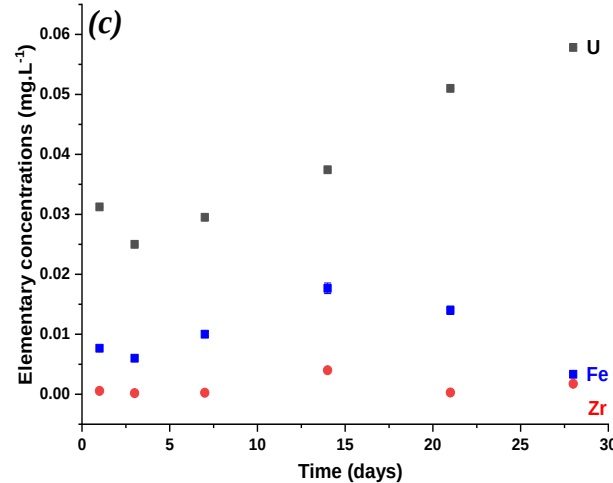
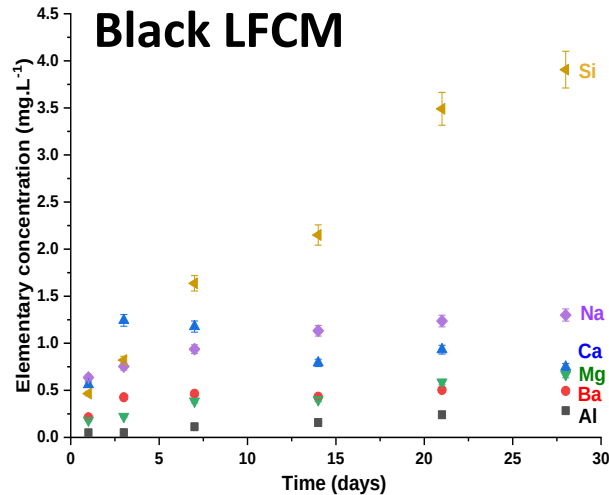


Corrosion of Chernobyl LFCM

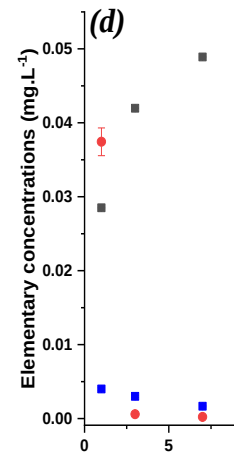
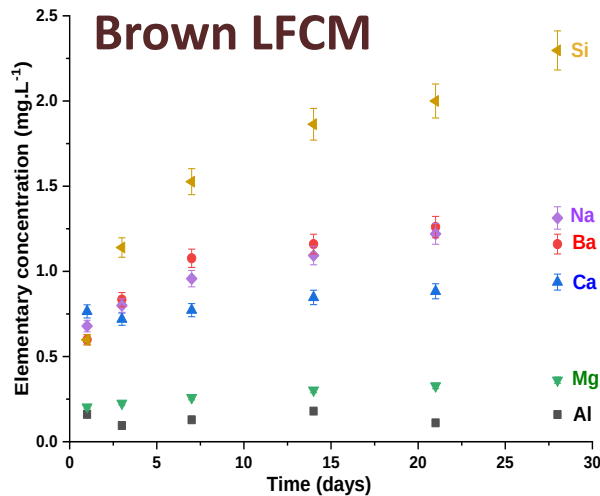


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**ASTM 1285 (PCT-B),
50°C, 35 days.
Analysis by ICP-MS.**



Black LFCM: $NR_U = (9.28 \pm 1.19) \times 10^{-5} \text{ g m}^{-2} \text{ d}^{-1}$

Brown LFCM: $NR_U = (7.28 \pm 1.51) \times 10^{-5} \text{ g m}^{-2} \text{ d}^{-1}$

Real Black LFCM: $NR_U = 1.8 \times 10^{-5} \text{ g m}^{-2} \text{ d}^{-1}$

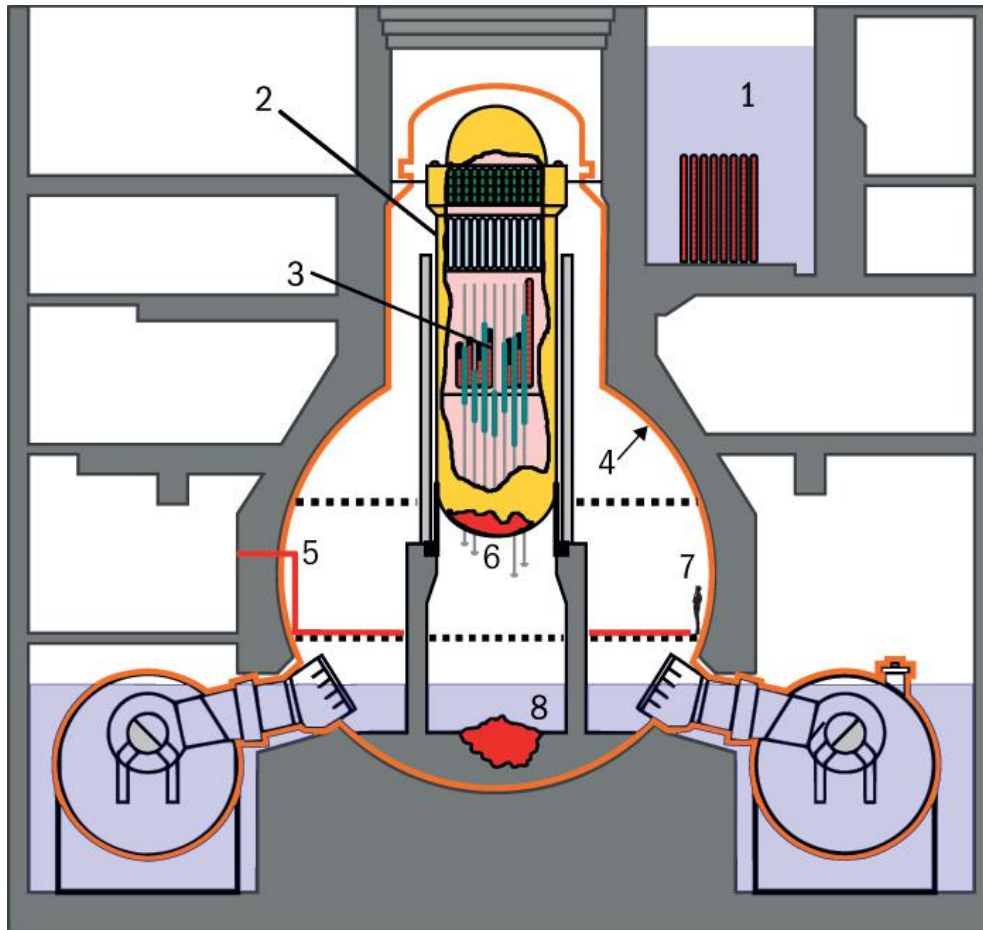
Rogozin et al. Radiochemistry, 4, 160 (1991)

Fukushima Daiichi fuel debris



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- | | |
|-------------------------------------|---|
| 1: Spent fuel pool | 5: Path of Hitachi shape-changing robot |
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| 3: Damaged core assembly | 7: Person for scale |
| 4: Primary containment vessel (PCV) | 8: Fuel debris/corium on basement floor |

Fuel retrieval due to start in the next few years: Concerns over Pu inventory of fuel debris



Synthesis of $\text{U}_{(1-x-y)}\text{Zr}_x\text{Ce}_y\text{O}_2$



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Room Temperature
5 M NH_4OH – pH = 9

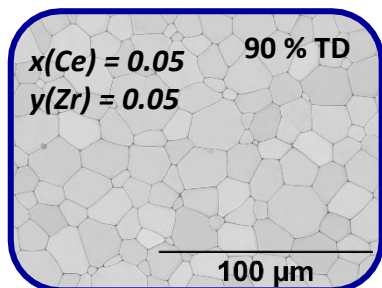
U (IV)
Zr(IV) ; Ce(III)

Drying
T : 90°C

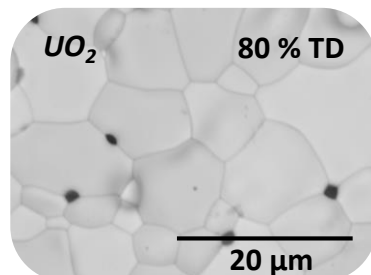
Washing steps
(water + ethanol)

Sintering

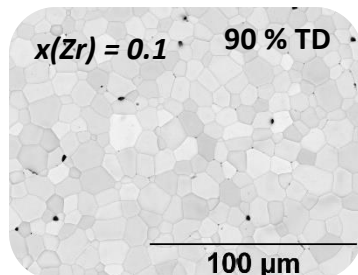
1700°C
8 h – N_2/H_2



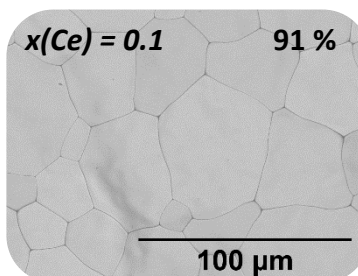
Grain size: $13.2 \pm 0.5 \mu\text{m}$



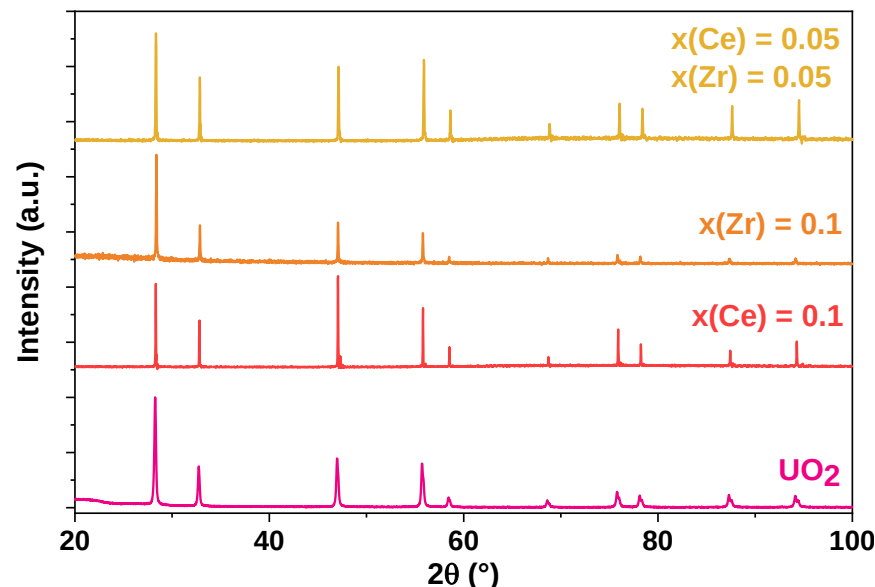
Grain size: $5.3 \pm 0.4 \mu\text{m}$



Grain size: $8.3 \pm 0.4 \mu\text{m}$



Grain size: $27.4 \pm 0.5 \mu\text{m}$



Composition	Lattice parameter (Å)	Literature
UO_2	5.467 ± 0.002	5.468 ± 0.001
$\text{U}_{0.90}\text{Ce}_{0.10}\text{O}_{2\pm y}$	5.462 ± 0.001	5.4647
$\text{U}_{0.90}\text{Zr}_{0.07}\text{O}_{2+y}$	5.437 ± 0.001	5.429 ± 0.002
$\text{U}_{0.90}\text{Ce}_{0.06}\text{Zr}_{0.03}\text{O}_{2\pm y}$	5.436 ± 0.001	—

- ✓ Incorporation of Ce & Zr into fluorite structure
 - Lattice contraction due to the incorporation of smaller size Zr^{4+} in UO_2

$\text{U}^{4+}(\text{VIII}): 1 \text{ Å}$ $\text{Zr}^{4+}(\text{VIII}): 0.84 \text{ Å}$ $\text{Ce}^{3+}(\text{VIII}): 1.143 \text{ Å}$

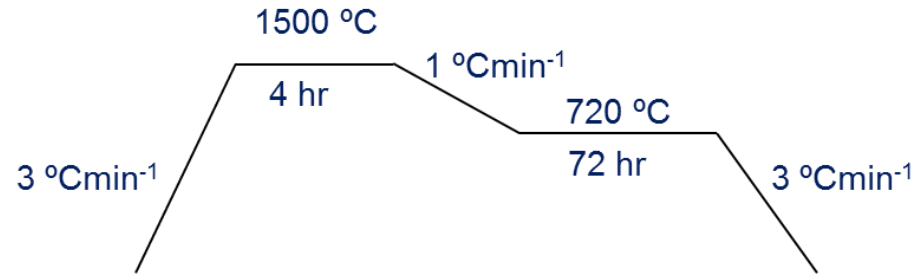
Desgranges, *Inorg. Chem.*, 2009
Kulkarni, *J. Nucl. Mater.*, 2009
Norris, *J. Nucl. Mater.*, 1983

Synthesis of MCCI



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Component	Wt.%
SiO ₂	43.6
CaO	8.85
Al ₂ O ₃	10.78
ZrO ₂	4.56
Stainless Steel	2
U _{0.9} Ce _{0.05} Zr _{0.05} O ₂	10.87

} Fukushima cement mortar

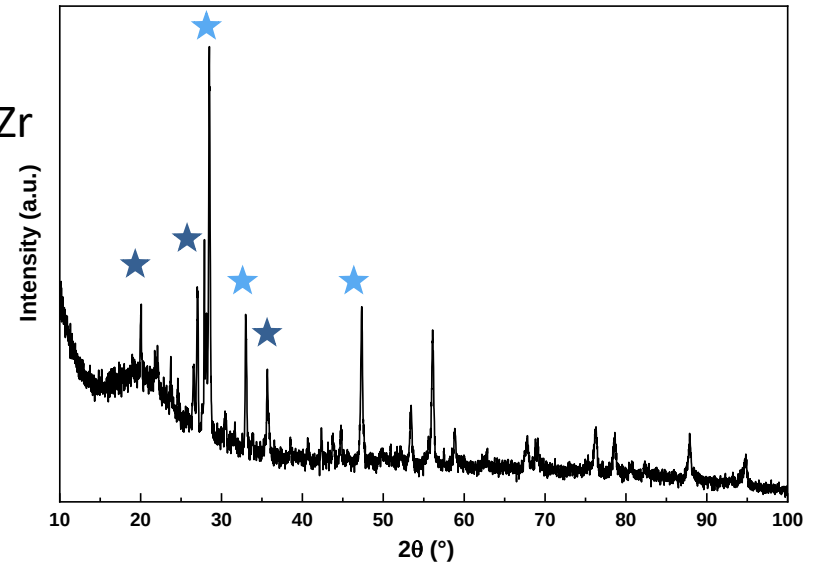
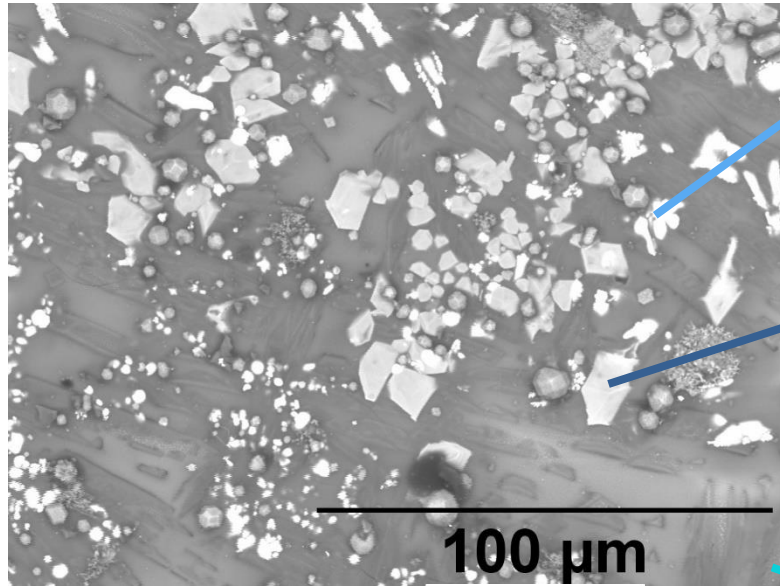


Characterisation of MCCI



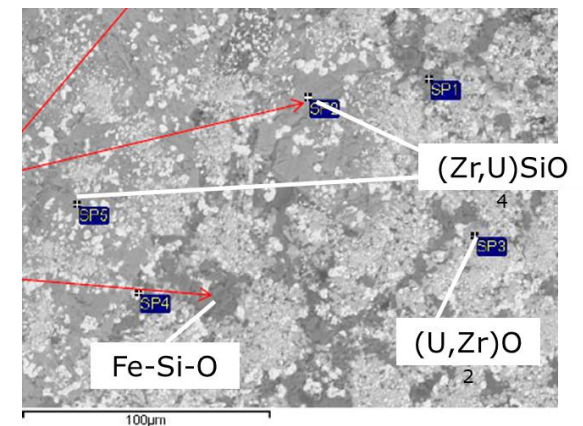
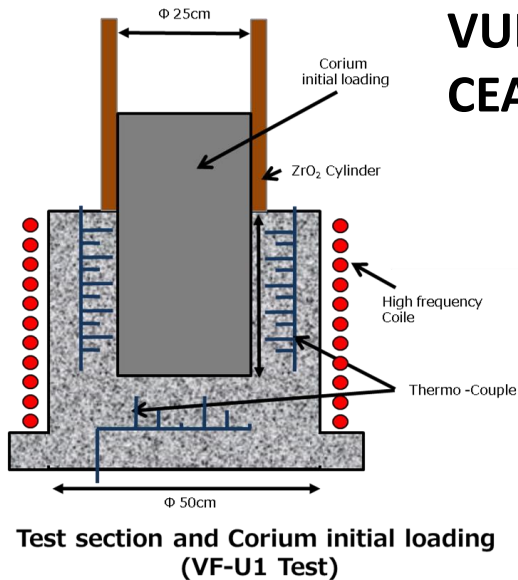
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Glass matrix

**VULCANO test facility,
CEA**



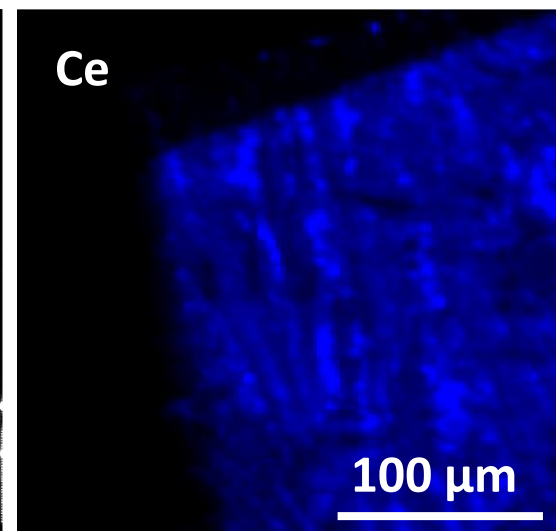
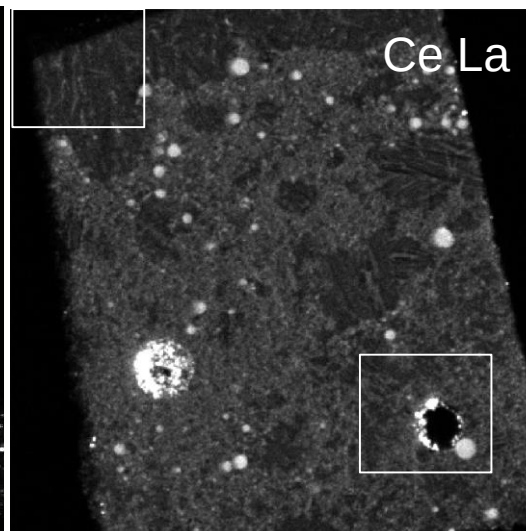
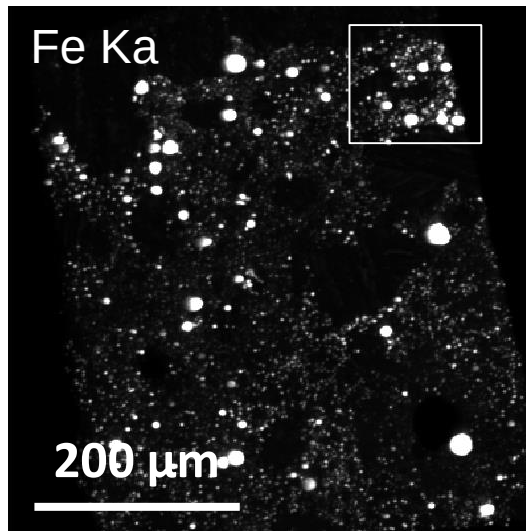
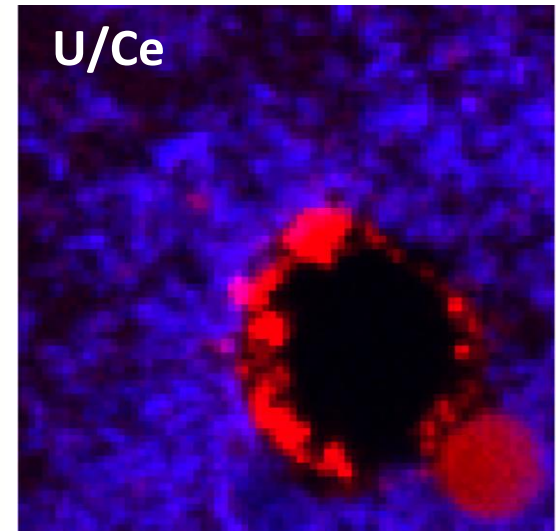
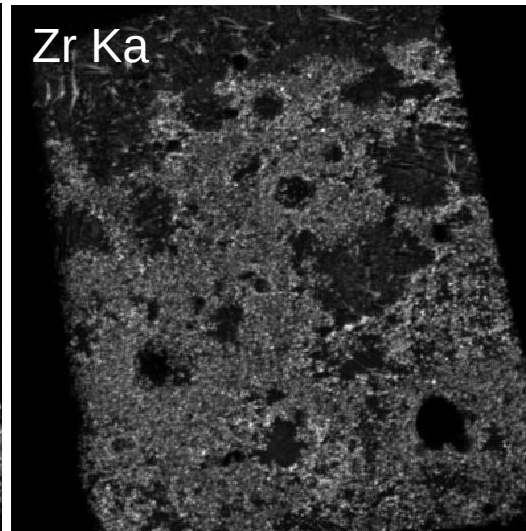
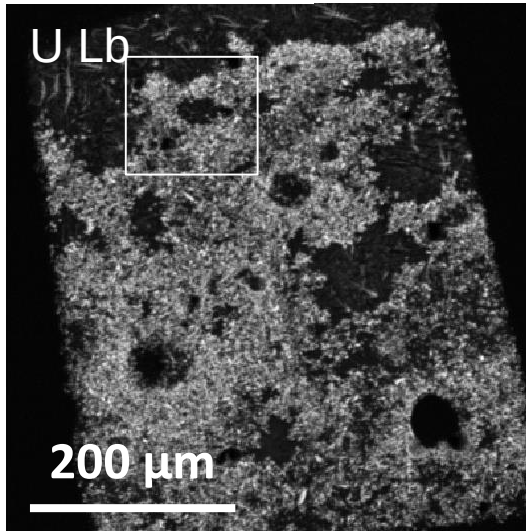
Characterisation of MCCI

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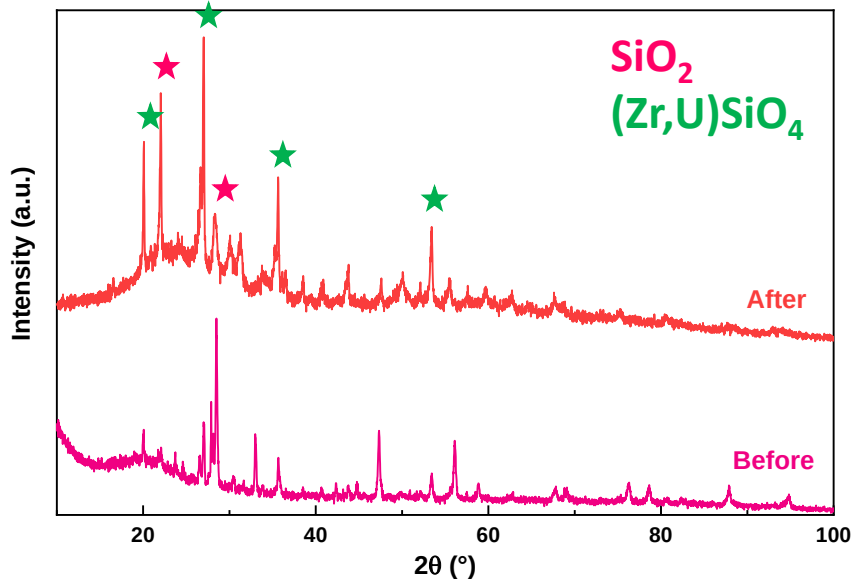
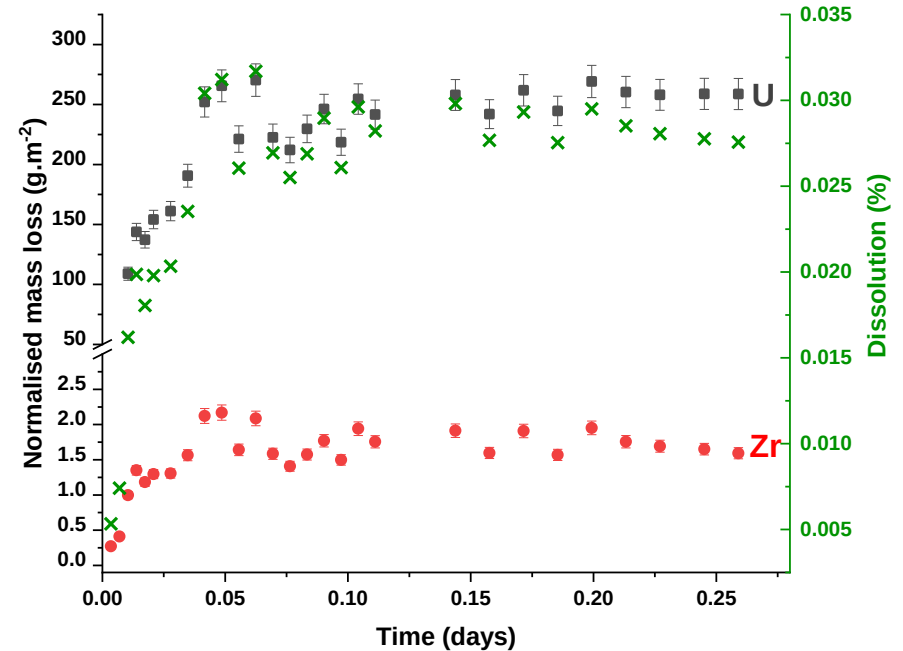
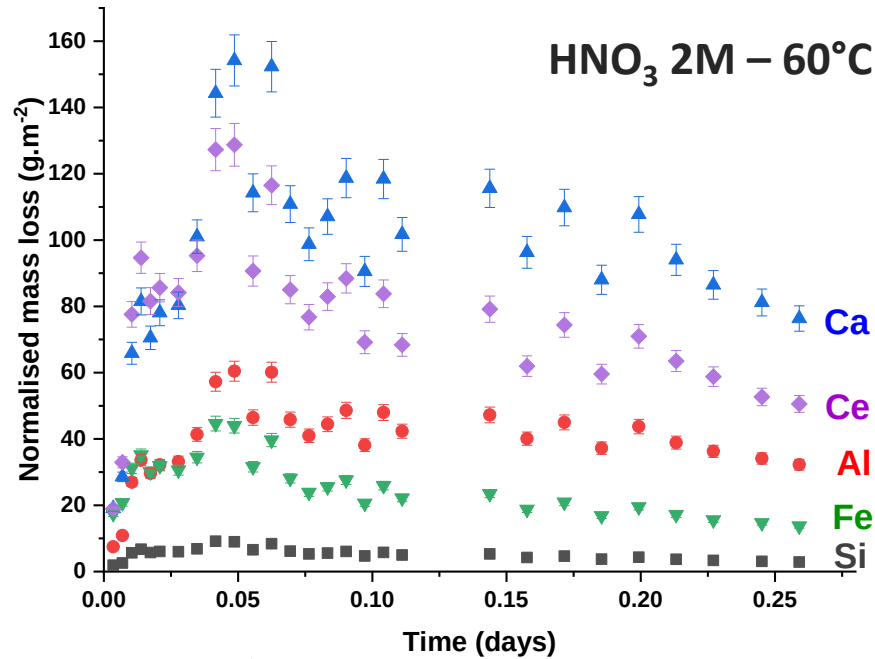


Corrosion behaviour of MCCI



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- UO₂ and ZrO₂-phases decreased
- Increased amorphous content
- U and Zr-phases reached steady state (passivating layer?)
- Ce follows behavior of glass elements

Conclusions



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Decommissioning nuclear accident reactors requires an understanding of the complex nature of the fuel debris

We have developed a suitable low-radioactivity surrogate for nuclear fuel that bears good resemblance to Chernobyl “lavas”

We have further developed these materials for application to Fukushima decommissioning, particularly to understand behaviour of cerium (plutonium)

Ongoing experiments are trying to understand the corrosion / dissolution behaviour of nuclear lavas and meltdown minerals

We are working with partners in Ukraine and Japan to help implement successful and timely decommissioning.

Thank you for your attention!

